MONROE TWP MUNICIPAL WELLS #4 & #5 MONROE TWP/GLOUCESTER COUNTY NEW JERSEY EPA # NJD980769699

A program for sampling raw water sources throughout the state revealed mercury contamination in Monroe Municipal Wells #4 & #5 on June 21, 1976. The levels of mercury that were in the municipal water system were as high as 10.8 ppb. This level is far in excess of the maximum contaminant level for total mercury of 2.0 ppb promulgated by the USEPA pursuant to the Safe Drinking Water Act.

These wells are underlain by the Quarternary Bridgeton Formation which is characterized by gravel and sand in part solidified by iron oxide. The water bearing unit in which Well #4 & #5 are located is the Cohansey Aquifer. The depth of Well #4 and #5 are 106 and 160 feet respectively. There are also two surface water swalls leading to the Squankum Branch and the Hospitality Branch of Great Egg Harbor River.

The sampling and analysis of an individual domestic well owned by the Monroe Township Board of Education revealed total mercury as high as 3.6 ppb on July 17, 1986. This well is located in a maintenance building adjacent to the Williamstown High School on Clayton Avenue in Williamstown. This well is also located at the South-West limit of the MTMUA water supply system.

Immediately after the discovery of mercury contamination in the public water system, Well #4 was taken out of service and was retained only for emergency situations. Well #5 was kept in service because levels of mercury experienced at this well were at or slightly below the maximum contaminant level. Since that time the level of mercury has steadily decreased in both wells.

At present it is not known conclusively where the source of the contamination is located. However, there are two potential sources. The first potential source is the Williamstown Substation of the Atlantic City Electric Company. This site is located adjacent to the high school property on Clayton Road. The second, and more probable source of contamination is an inactive gravel pit also located on Clayton Road. This site has had unauthorized dumping of assorted solid waste over many years since its closure.

It is recommended that this site be referred to NJDEP/Division of Water Resources for a comprehensive sampling and analysis of the ground water in Williamstown. This will provide additional information to determine the extent and source of the contamination. All environmental sampling undertaken by the NJDEP should be closely coordinated and integrated with the Glouester County Health Department which is also aware of this problem.

Submitted by:

Frank Faranca, HSMS IV NJDEP/DHWM/BPA MSCA Project

Hours worked: 47

FF:mz

;÷



Preliminary Assessment

Monroe Township Municipal Wells #4 & #5

#4 - Washington Avenue

#5 - Chestnut & Water Streets

Monroe Township/Gloucester County

New Jersey

EPA #NJD980769699

SEPA

POTENTIAL HAZARDOUS WASTE SITE PRELIMINARY ASSESSMENT PART 1 - SITE INFORMATION AND ASSESSMENT

L IDENTIFICATION

O1 STATE 02 SITE NUMBER

NJ D9807 69699

WEITA	PART 1 - SIT	E INFORMAT	ON AND	ASSESSME	TNJ I	980/69699
II, SITE NAME AND LOCATION						
O1 SITE NAME Kagas, sommon, or decompany name of same		(2 STREET, A	OUTE NO., OR	SPECIFIC LOCATION IDENTIFIER	
Monroe Township Municipal	l Wells 🕯				e; Chestnut & Wat	
03 CTY) c		ZIP CODE		07 COUNTY 08 CONG CODE DIST
Monroe Township		L	NJ C	08094	Gloucester	08 DIST
09 COORDINATES LATITUDE	LONGITU	DE	Well #	4: Blo	ck 3202, Lot 17	·
	<u>5 03 40 .</u>				ck 11602, Lot 2	
From Trenton, take Route Williamstown.	295 Sout	h to Rout	e #42.	Follo	w Route #42 Sout1	n to
III. RESPONSIBLE PARTIES						
01 OWNER is enoug		7	2 STREET	lutnoss, manng, /s	16-Dent-eli	
Monroe Twp. M.U.A.					n Street	
O3 CITY		10	A STATE 05	ZIP CODE	06 TELEPHONE NUMBER	George
Williamstown		1	U [08094	(609) 629-7586	Cassabone
07 OPERATOR (F aroun and sciolant from owner)			STREET IS		I Andrews	
			0.02.00			· · · · · · · · · · · · · · · · · · ·
O9 CITY			O STATE 11	ZIP CODE	12 TELEPHONE NUMBER	1
13 TYPE OF OWNERSHIP (Crocs and)					<u></u>	<u> </u>
□ A. PRIVATE □ B. FEDERAL:	·	(Agency name)			E DD.COUNTY DE. MU	NICIPAL
	/Scieces			Ci G. UNKA	HOWN	
THE DIVING MICHAEL SEASON OF THE ICANON	-					_
_ A. RCRA 3001 DATE RECEIVED:	DAY YEAR	. UNCONTROLLE	D WASTE S	ITE (CERCLA 10:	DATE RECEIVED: 1	T YEAR C. NONE
IV. CHARACTERIZATION OF POTENTIAL H	MAZARD					
OT ON SITE INSPECTION OF YES DATE 2 8 77 CI NO MONTH DAY YEAR	BY (Chack of) A. EPA E. LOCA	O B. EPA (L HEALTH OFFIC				CONTRACTOR
	CONTRACT	FOR NAME(S): _			(Spealy)	
02 SITE STATUS (Crees ene)	03	YEARS OF OPERA	TION			
A. ACTIVE B. INACTIVE C. UNI	KNOWN	-30	1951			N
04 DESCRIPTION OF SUBSTANCES POSSIBLY PRESE	NT. KNOWN, OR A					
Inorganic mercury						
inorganize mereally						
05 DESCRIPTION OF POTENTIAL HAZARD TO ENVIRON	MMENT AND OR	OBIH ATION		 		
Inorganic mercury was de			รแกกไซ	wells	#4 and #5 as high	as 6.4 ppb
7/10/76 Was de	alaa Ja	rostod d-		abla wa	ll at the limit o	f the
			a pota	abie we.	ii at the iimit o	1 the
Monroe MUA Public Water	Supply S	ystem.				•
V. PRIORITY ASSESSMENT	 					
01 PRIORITY FOR INSPECTION (Choca one. # non or mean	m & Checked, Campie	is Parl 2 - Wasie Inform	euron and Peri 3	- Description of Ma	reroous Conducts and Incounts)	
☐ A. HIGH ☐ B. MEDIL		C. LOW		D. NON		Laten formu
VI. INFORMATION AVAILABLE FROM						
01 CONTACT	1	OF (Agency/Organizati	an)			03 TELEPHONE NUMBER
George Cassabone, Superi	ntendent	Monroe '	Iwp. M	.U.A.		(609) 629-758
C4 PERSON RESPONSIBLE FOR ASSESSMENT	05	AGENCY	06 ORGANI		07 TELEPHONE NUMBER	08 DATE
Frank Faranca, HSMS IV	N.	JDEP ·	DHWM/	BPA	(609) 633-2219	04 14 87

WEPA	V		•	4
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POTENTIAL HAZARDOUS WASTE SITE

L IDENTIFICATION

VEI	A		PRELIMINARY PART 2 - WASTE	ASSESSMENT EINFORMATION		NJ D980	769699
II. WASTE S	TATES, QUANTITIES, AN	ID CHARACTERI	STICS				
LI A SOLIO	CAL STATES (Chock of the supply) OLID L2 E. SLURRY OWDER, FINES L1 F LIQUID LUGGE L1 G GAS CUBIC YARDS		LI A TOXIC LI E SOLUBLE LI I HIGHLY VOLA LI B CORROSIVE LI F. INFECTIOUS LI J. EXPLOSIVE LI C. RADIOACTIVE LI G. FLAMMABLE LI L. INCOMPATIBLE LI D. PERSISTENT LI HI IQUITABLE LI L. INCOMPATIBLE LI M. NOTABLE LI L. INCOMPATIBLE LI M. NOTABLE LI L. INCOMPATIBLE LI M. NOTABLE LI M. NOTA				VE /E ATIOLE
III. WASTE T				<u> </u>			
CATEGORY	SUBSTANCE N	IAME	01 GROSS AMOUNT	02 UNIT OF MEASURE	03 COMMENTS		
SLU	SLUDGE						
OLW	OILY WASTE						
SOL	SOLVENTS						
PSO	PESTICIDES						
೦೦೦	OTHER ORGANIC ÇI	1EMICALS					
100	INORGANIC CHEMIC	ALS				s detected in	
ACD	ACIOS				water syst	em as high as	s 10.8 ppb
BAS	BASES						
MES	HEAVY METALS		Unknown	l	<u></u>	<u> </u>	
	OUS SUBSTANCES			r			DA MEASURE DE
01 CATEGORY	02 SUBSTANCE N	AME	03 CAS NUMBER	04 STORAGE DIS	OSAL METHOD	05 CONCENTRATION	06 MEASURE OF CONCENTRATION
MES	Mercury		7439-97-6	Unknown		10.8	PPB
	<u> </u>		 				
	·		!			! 	!
 				 			
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 -				<u> </u>			
			 				
					· 		
V SEEDSTO	incre .		<u> </u>	<u> </u>		<u> </u>	
	CKS :See Appendix No CAS Numb						
CATEGORY	O1 FEEDSTOO	A NAME	02 CAS NUMBER	CATEGORY	O1 FEEDSTO	OCR NAME	02 CAS NUMBER
FDS			ļ	FDS		·	
FDS				FDS			
FDS FDS		· · · · · · · · · · · · · · · · · · ·	<u> </u>	FOS			
	5 OF INFORMATION :		1	FOS		<u> </u>	
VI. SOURCE	S OF INFORMATION ICA	turcus inforticas, e y.	state laus, sandie anarysis, i	/000/16)	·		
See ref	erence sheet f	or list of	attachment	s.			

POTENTIAL HAZARDOUS WASTE SITE

L IDENTIFICATION O1 STATE Q2 SITE MANGER NJ D980769699

PRELIMINARY ASSESSMENT
PART 3 - DESCRIPTION OF HAZARDOUS CONDITIONS AND INCIDENTS

IL HAZARDOUS CONDITIONS AND INCIDENTS	
01 L. A. GROUNDWATER CONTAMINATION 8254	02 0 OBSERVED (DATE: 7-19-76) D POTENTIAL C ALLEGED
US POPULATION POTENTIALLY AFFECTED.	04 NARRATIVE DESCRIPTION
	sampled in 1976 and mercury was detected as
	potable well was found to contain 3.6 ppb on
7/17/86.	Attachment B3, B6 and C
01 L.B. SURFACE WATER CONTAMINATION	02 LOBSERVED (DATE) LO POTENTIAL LE ALLEGED
03 POPULATION POTENTIALLY AFFECTED:	04 NARRATIVE DESCRIPTION
	ay have contributed to surface water contamina- e Hospitality Branch of the Great Egg Harbor
River.	e hospitality branch of the Great Egg harbor
	Map 1 .
01 I.C. CONTAMINATION OF AIR 03 POPULATION POTENTIALLY AFFECTED:	02 C OBSERVED (DATE) C) POTENTIAL C: ALLEGED 04 NARRATIVE DESCRIPTION
	y in the ground water to contaminate the air.
no potential exists for the mercury	y in the ground water to contaminate the all.
01 (D FIRE/EXPLOSIVE CONDITIONS	02 C: OBSERVED (DATE:) [] POTENTIAL [] ALEGED
03 POPULATION POTENTIALLY AFFECTED:	04 NARRATIVE DESCRIPTION
	xplosive conditions due to the nature of the
contaminant.	·
S3 POPULATION POTENTIALLY AFFECTED	OF LASSIBLE OF THE COLOR POTENTIAL ALEGED OF NARRATIVE DESCRIPTION
No potential exists for direct cont	act with the mercury contaminated ground water.
01 C F CONTAMINATION OF SOIL 03 AREA POTENTIALLY AFFECTED:	02 O OBSERVED (DATE:) [] POTENTIAL
(ACIOL)	exists due to the documented mercury contamina-
tion found in the ground water.	exists due to the documented mercury contamina-
cross todied in the Stodied water.	Attachment B & C
	·
01 L. G. DRINKING WATER CONTAMINATION 03 POPULATION POTENTIALLY AFFECTED.	02 [] OBSERVED (DATE:) [] POTENTIAL [] ALLEGED 04 NARRATIVE DESCRIPTION
Public supply wells #4 and #5 were	sampled and analysis revealed inorganic mercury
	er standards. Well #4 is currently not in use.
	Attachment B1-B6
01 C H. WORKER EXPOSURE/INJURY	02 D OBSERVED IDATE D POTENTIAL D ALLEGED
03 WORKERS POTENTIALLY AFFECTED:	04 NARRATIVE DESCRIPTION
A potential exists for worker expos	sure/injury because well #4 is utilized in
emergency situations.	
	Attachment B, C
01 G I. POPULATION EXPOSURE/INJURY	02 (1 OBSERVED (DATE) D POTENTIAL D ALLEGED
03 POPULATION POTENTIALLY AFFECTED: 8254	04 NARRATIVE DESCRIPTION
Approximately 32% of Monroe Townshi	ip is not on public water supply or 8254 people.
	. I september 1
	Attachment H
•	ACC actiment in

SEPA

POTENTIAL HAZARDOUS WASTE SITE PRELIMINARY ASSESSMENT

L IDENTIFICATION

O1 STATE 02 SITE NUMBER

NJ D980769699

PART 3 - DESCRIPTI	ON OF HAZARDOUS CONDITIONS AND INCIDENTS	LNO LDAG	507 0 9 0 9 9
II. HAZARDOUS CONDITIONS AND INCIDENTS	Comment :		
01 J. DAMAGE TO FLORA 04 NARRATIVE DESCRIPTION	02 OBSERVED (DATE:)	D POTENTIAL	□ ALLEGED
Surface water drainage and g	ground water discharge may potenti	ally impact	aquatic
and terrestrial flora with r	•	lap 1 '	_
01 T K DAMAGE TO FAUNA 04 NARRATIVE DESCRIPTION (MEANS NAME) OF ADDRESS	02 C OBSERVED (DATE:)	C POTENTIAL	C ALLEGED
Surface water drainage and g	ground water discharge may potenti	ally impact	aquatic
and terrestrial fauna with m	nercury contamination.	Map 1	
01 T L CONTAMINATION OF FOOD CHAIN	02 OBSERVED (DATE)	2 POTENTIAL	O ALLEGED
O4 NARRATIVE DESCRIPTION The mode of action of mercus	y in organisms is that it is Lipi	o soluable t	hus
	for contaminating the food chain.		
making it a potential agent		ttachment G	
01 C M UNSTABLE CONTAINMENT OF WASTES	02 () OBSERVED (DATE:)	POTENTIAL	C ALLEGED
C3 POPULATION POTENTIALLY AFFECTED:	04 NARRATIVE DESCRIPTION	:	
	able mercury waste to be contamina	ting the gro	ound & C
Water. 01 C N DAMAGE TO OFFSITE PROPERTY	02 OBSERVED (DATE:)	[] POTENTIAL	D ALLEGED
04 NARAATIVE DESCRIPTION			- T1
Analysis of a potable well of	owned by the Monroe Twp. Board of 5 ppb which is in excess of the ma	Education in	n July minant
level for safe drinking water		ttachment C	miname
	UNS. WWTPs 02 OBSERVED (DATE:)	C POTENTIAL	D ALLEGED
	anding water in a storm sewer culv	ert which a	ccepts
well #4 overflow revealed me	ercury below detectable limits.	Attachment B	_30
01 C P ILLEGAL/UNAUTHORIZED DUMPING	02 □ OBSERVED (DATE:)	D POTENTIAL	C) ALLEGED
04 NARRATIVE DESCRIPTION	· · · · · · · · · · · · · · · · · ·		
-	horized dumping of mercury waste	to be contr	louting
to the present ground water	CONLAIMINACTON.		
C5 DESCRIPTION OF ANY OTHER KNOWN, POTENTS	AL. OR ALLEGED HAZARDS		
	re not hooked up to the Municipal	Water Suppl	y and are
at risk of being exposed to	the contaminated ground water thr	ough the us	e of
their domestic wells.			
III. TOTAL POPULATION POTENTIALLY AFFECT	EO:		
IV. COMMENTS			
Sampling and analysis of por	table water from the surrounding h	nomes along	Clavton
and Janvier Roads is recomme		,	
	•		
V. SOURCES OF INFORMATION (Con LOCALE Information	I. 6 g , Listo Ires, Lemon energial, reportu		
See reference sheet for list	t of attachments.		
			-
	•		

Dangerous Properties of Industrial Materials

Sixth Edition

N. IRVING SAX

Assisted by:

Benjamin Feiner/Joseph J. Fitzgerald/Thomas J. Haley/Elizabeth K. Weisburger

TOXICITY DATA: '3 CODEN:

Aquatic Toxicity Rating: TLm96:under 1 ppm WQCHM* 3,-,74. DOT: Poison B, Label: Poison FEREAC 41,57018,76. Occupational Exposure to Inorganic Mercury reem std: Air: TWA 0.05 mg(Hg)/m3 NTIS**. Reported in EPA TSCA Inventory, 1980.

THR: A poison. See also mercury compounds and bromides.

Disaster Huzard: When heated to decomp it emits very tox fumes of Br⁻ and Hg. Incomp: Indium, Na, K.

MERCURY(II) BROMIDE complex with TRIS(2-ETHYLHEXYL) PHOSPHITE

CAS RN: 64011376 NIOSH #: TH 2275000 mf: C₂₄H₅₁O₃P+Br₂Hg; mw: 779.13

SYN: PHOSPHOROUS ACID, TRIS(2-ETHYLHEXYL) ESTER, COMPLEX WITH MERCURY(II) BROMIDE (1:1)

TOXICITY DATA: 3 CODEN: ipr-mus LDLo: 31 mg/kg CBCCT* 7,790,55

Occupational Exposure to Inorganic Mercury recm std: Air: TWA 0.05 mg(Hg)/m3 NTIS**.

THR: HIGH ipr. See also mercury compounds and bromides.

Disaster Hazard: When heated to decomp it emits very tox fumes of PO_x, Br⁻ and Hg.

MERCURY(II) CHLORIDE complex with TRIS(2-ETHYLHEXYL) PHOSPHITE

CAS RN: 63981497 • NIOSH #: TH 2450000 mf: C₂₄H₅₁O₃P•Cl₂Hg; mw: 690.21

SYN: PHOSPHOROUS ACID, TRIS(2-ETHYLHEXYL) ESTER, COMPLEX WITH MERCURY(II) CHLORIDE (1:1)

TOXICITY DATA: 3 CODEN: ipr-mus LDLo:03 mg/kg CBCCT* 7,791,55

Occupational Exposure to Inorganic Mercury recm std: Air: TWA 0.05 mg(Hg)/m3 NTIS**.

THR: HIGH ipr. See also mercury compounds.

Disaster Hazard: When heated to decomp it emits very tox fumes of Hg, Cl⁻ and PO_x.

MERCURY(I) CHLORITE

mf: ClHgO₂; mw: 268.04

THR: Unstable when dry (explodes spontaneously). A poison. See also mercury compounds, chlorides. Disaster Hazard: When heated to decomp it emits very tox furnes of Hg and Cl⁻.

MERCURY COMPOUNDS, INORGANIC

THR: Mercury is a general protoplasmic poison; after absorption it circulates in the blood and is stored in the liver, kidneys, spleen and bone. It is eliminated in the urine, feces, sweat, saliva and milk. In industrial poisoning, the principal effect is upon the CNS and upon the mouth and gums. Colitis has been reported frequently; a nephritis or nephrosis is rarely reported.

Fulminate of mercury rarely produces symptoms of systemic poisoning, but frequently causes a dermatitis. The cardinal symptoms of industrial mercury poisoning are stomatitis, tremors, and psychic disturbances. Usually the first complaints are of excessive salivation and pain on chewing; in severe cases there may be gingivitis, with loosening of the teeth, and a dark line on the gum margins, resembling the "lead line." In slow poisoning the salivation may be absent, and the only complaint dryness of the throat and mouth. Tremor and psychic disturbances are commonly seen in the slow chronic form of the poisoning; the tremor is of the intention type, and may be seen when the patient spreads the outstretched fingers or protrudes the tongue, or attempts to perform specified movements. Muscles of the face, hands and arms are chiefly affected. In more severe cases there may also be convulsive or shaking movements; writing is frequently illegible. Hyperactive kneejerks and scanning speech may be present in advanced cases. The psychic disturbance (so called "erethism") includes such changes as loss of memory, insomnia, lack of confidence, irritability, vague fears and depression.

The dermatitis produced by fulminate of mercury takes the form of small, discrete ulcers on the exposed parts; and is usually accompanied by conjunctivitis and inflammation of the mu mem of the nose and throat. In hmns it is readily absorbed via respiratory tract (elemental mercury vapor, mercury composed dusts) intact skn, and G.I. tract, although occasional incidental swallowing of metallic Hg is without harm. Spilled and heated elemental Hg is particularly hazardous. A number of mercury compounds, in addition to the fulminate, can cause skn irr and be absorbed through the skn. They are strong allergins; common air contaminants

Acute Tox: Sol salts have violent corrosive effects on skn and mu mem; severe nausea, vomiting, abdominal pain, bloody diarrhea, kidney damage; death usually within 10 days.

Disaster Hazard: Dangerous; when heated to decomp emits tox fumes of Hg.

MERCURY COMPOUNDS, ORGANIC

THR: The customary grouping of all organic mercurials in a single category is not fully justified by the toxicity of the compounds. Alkyl mercurials have very high toxicity; aryl compounds, particularly the phenyls, are much less toxic, and the organomercurials used in therapeutics are less toxic. The alkyls and aryls commonly cause skn burns and other forms of irr, and both can be absorbed through the skn. Fatal poisoning has occurred due to exposure to alkyl mercurials and permanent damage to the brain has been reported. Extensive human observation on exposure to the phenyl mercurials have shown no greater toxicity than is caused by metallic mercury. In fact, up to the present time there has not been an authenticated case of occupational poisoning due to the phenyl mercurials reported in the

THE MERCK INDEX

AN ENCYCLOPEDIA OF CHEMICALS AND DRUGS

NINTH EDITION

ELSAMDY

Martha Windholz, Editor
Susan Budavari, Associate Editor
Lorraine Y. Stroumtsos, Assistant Editor
Margaret Noether Fertig, Assistant Editor

Published by

MERCK & CO., INC.

RAHWAY, N.J., U.S.A.

1976

ATTACHMENT F

Hg; at. wt 200.59; at. no. 80; valences 1, 2. Group 2b element. Abundance in earth's crust 0.5 ppm. Natural iso-

topes: 202 (29.80%); 200 (23.13%); 199 (16.84%); 201 (13.22%); 198 (10.02%); 204 (6.85%); 196 (0.146%); known isotopes range in mass number from 189 to 206. Obtained by roasting cinnabar (mercuric sulfide). Reviews: Roberts, Advan. Inorg. Chem. Radiochem. 11 (Academic Press, New York, 1968) pp 309-339; Aylett, "Group IIB" in Comprehensive Inorganic Chemistry vol. 3 (Pergamon Press, Oxford,

1973) pp 187-328.

Silver-white, heavy, mobile, liquid metal; slightly volatile at ordinary temp; solid mercury is a tin-white, ductile, malleable mass which may be cut with a knife. mp -38.87°; bp 356.72°; d²⁵ 13.534; C₂₅ (25°) 6.687 cal/mole deg. Vapor pressure (25°): 2×10^{-3} mm; heat of vaporization (25°): 14.652 kcal/mole: Busey, Giauque, J. Am. Chem. Soc. 75, 806 (1953). Surface tension (25°): 484 dynes/cm; electrical resistivity (20'): 95.76 µohm cm. When pure does not tarnish on exposure to air at ordinary temp, but when heated to near the boiling point slowly oxidizes to HgO. Forms alloys with most metals except iron and combines with sulfur at Fordinary temp. E^0 (aq) $Hg/Hg^2+ -0.854$ V; E^0 (aq) 2 $Hg/Hg_2^{2+} -0.789 \text{ V. Soly in water (25°): } 0.28 \ \mu\text{moles/l;}$ data on soly in organic solvents: Spencer, Voigt, J. Phys. Chem. 72, 464 (1968). Reacts with HNO, and hot, concd H₂SO₄; does not react with dil HCl, cold H₂SO₄, or alkalies. Reacts with ammonia solns in air to form Hg, NOH, Millon's base. Mercury salts when heated with Na₂CO₃ yield metallic Hg and are reduced to metal by H₂O₂ in the presence of alkali hydroxide. Cu, Fe, Zn and many other metals ppt metallic Hg from neutral or slightly acid solns of mercury salts. Soluble ionized mercuric salts give a yellow ppt of HgO with NaOH and a red ppt of HgI₂ with alkali iodide. Mercurous salts give a black ppt with alkali hydroxides and a white ppt of calomel with HCl or sol chlorides. They are slowly dec by sunlight. Poisonous!

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USE: In barometers, thermometers, hydrometers, pyrometers; in mercury arc lamps producing ultraviolet rays; in switches, fluorescent lamps; in mercury boilers; manuf all mercury salts, mirrors; as catalyst in oxidation of organic compds; extracting gold and silver from ores; making amalgams, electric rectifiers, mercury fulminate; also in dentistry; in determining N by Kjeldahl method, for Millon's reagent; as cathode in electrolysis, electroanalysis, and many other. uses. Also in pharmaceuticals, agricultural chemicals, antifouling paints. Human Toxicity: Readily absorbed via respiratory tract (elemental mercury vapor, mercury compd dusts), intact skin, and G.I. tract, although occasional incidental swallowing of metallic mercury is without harm. Spilled and heated elemental mercury is particularly hazardous. Acute: sol salts have violent corrosive effects on skin and mucous membranes; severe nausea, vomiting, abdominal pain, bloody diarrhea; kidney damage; death usually within 10 days. Chronic: inflammation of mouth and gums, excessive salivation, loosening of teeth; kidney damage; muscle tremors, jerky gait, spasms of extremities; personality changes, depression, irritability, nervousness. Phenyl and alkyl mercurials can cause skin burns and be absorbed by the skin. Burning sensation is delayed several hours and thus gives no warning. Alkyls have affinity for brain tissue and may cause permanent damage. Phenyls are no more toxic than inorganic Hg. Antidole. Dimercaprol (BAL). See E. Browning, Toxicity of Industrial Metals (Appleton-Century-Crofts, New York, 2nd ed., 1969) pp 226-242.

PESTICIDES IN THE ENVIRONMENT

Volume 1, Part I (In Two Parts)

Edited by

ROBERT WHITE-STEVENS

CHAIRMAN. BUREAU OF CONSERVATION AND ENVIRONMENTAL SCIENCE COLLEGE OF AGRICULTURE AND ENVIRONMENTAL SCIENCE RUTGERS UNIVERSITY—THE STATE UNIVERSITY NEW BRUNSWICK. NEW JERSEY

MARCEL DEKKER, INC., New York 1971

計劃條係

 .. METCALF

were intro-(139) when 195°C, and ly toxic to ndoned beounds that ichloro-4,6nzene, or 'CNB, m.p.

zene

CI NH₂ CI

.6-dichloro-4itroaniline

izene have id a newer , has been vegetables. °C, should it for smut

sis is lacksoil fungihloronitrois the most strachloro-1.5,6-tetraore active 1. CHEMISTRY AND BIOLOGY OF PESTICIDES

3. Mode of Action. Little critical work has been done on the mode of action of the chloronitrobenzenes. Brown(18) considers these compounds as vapor fungistats which, however, do not stop hyphal growth. Rich (117) states that pentachloronitrobenzene may be a competitive inhibitor of inositol, and essential growth substance for fungi.

4. Biological Activity. The chloronitrobenzenes are fungistats effective principally against soil fungi such as Botrytis, Rhizoctonia, Mucor, Trichoderma, and Fusarium. They act largely by fumigation and retard germination and colony growth as well as suppressing sporulation. These compounds are also useful in delaying sprouting of stored potatoes (19).

5. Toxicology. The oral LD_{50} values to the rat of dinitrotrichlorobenzene 500, and pentachloronitrobenzene of 1650 mg/kg suggest that these compounds are of low to moderate toxicity. Tetrachloronitrobenzene has been fed to rats at 100 mg/kg daily with no apparent ill effects, and rats survived two-year feeding studies with pentachloronitrobenzene at 2500 ppm.

G. Mercury Compounds

1. Introduction and Chemistry. The development of mercurial fungicides was an outgrowth of the usefulness of mercuric chloride as a bactericide. Mercuric chloride, HgClo. was first tested as a seed treatment on cereals by Kellerman and Swingle in 1890(71) and by Hiltner in 1915(59) who paved the way for the concept of protective seed dressings by observing that mercury treatment would prevent reinfestation by dormant mycelium of Fusarium disease of rye. However, the very poisonous nature of mercuric enforme prevented its widespread use until I. G. Farbenindustrie introduced an organic mercurial "chlorphenol mercury" CIC₆H₃OH·Hg·OSO₃Na in 1915(87), for use as a liquid seed disinfectant. From that time on, a succession of organic mercurials of varying chemical structures have been marketed as shown below. Organic mercury dusts for dry seed treatment were introduced in 1924 with onitrophenol mercury HOC₆H₃NO₂·HgOH. Dust treatments with organic mercurials are not only unpleasant but also are hazardous, and slurry treatments became popular with the introduction, about 1930, of a new type of organic mercurial, represented by methoxyethylmercury chloride CH3OCH2CH2Hg·Cl, developed for use as a slurry treatment or its corresponding silicate used as a dry seed treatment.

The organomercury fungicides can be represented by the general formula RHgX where R = aryl-, aryloxy-, alkyl-, or alkoxyethyl-, and X is an anionic group such as chloride, acetate, lactate, urea, carbamate, hydroxy, or related structure as shown in Table 1-3. The nature of the

TABLE 1-3

	Organomercury Fungicides	
Phenyl mercurials	•	
Phenyl mercuric acetate	$C_6H_3HgOCOCH_3$	Seed, turf, foliage, and industrial fungicide
Tolył mercuric acetate	CH ₃ C ₆ H ₄ HgOCOCH ₃	-
Phenyl mercuritriethanolammonium lactate	C ₆ H ₅ HgN(C ₂ H ₄ OH) ₃ ·OC(O)CH(OH)CH ₃	Scab treatment on fruits, ornamentals, turf
Phenyl mercurimonoethanolammonium acetate	$C_6H_5HgNH_2C_2H_4OH\cdot OC(O)CH_3$	Scab treatment on fruits, ornamentals
Phenyl mercury urea	$C_6H_5HgNHC(O)NH_2$	Seed treatment
Phenyl mercuric dimethyldithiocarbamate	$C_8H_5HgSC(S)N(CH_3)_2$	Slimicide for paper mills and mold retardant for paper
Phenoxy mercurials		
Hydroxy mercurichlorophenol	ClC ₆ H ₄ OHgOH	Seed, bulb, turf, foliage treatment
Hydroxy mercurinitrophenol	O ₂ NC ₆ H ₄ OH ₈ OH	Disinfectant for potatoes
Methyl mercury 8-hydroxyquinolate	OHgCH ₃	Seed fungicide
Ethyl mercurithiosalicylic acid	COONa SHgC ₂ H ₅	Seed and bulb fungicide
Alkyl mercurials	The second second	
Methoxyethylmercuric chloride	CH ₃ OCH ₃ CH ₃ HgCl	Seed treatment
Ethylmercuric chloride	C₂H₃HgCl	Seed treatment

Ethylmercuric acetate 1-Acetoxymercuri-2-hydroxyethane Methylmercuric dicyandiamide Ethylmercury 2,3-dihydroxypropyl mercaptide Chloromethoxypropylmercuric acetate

C₂H₃HgOC(O)CH₃ HOC₂H₄HgOC(O)CH₃ CH₃HgNHC(=NH)NHCN HOCH₂CH(OH)CH₂SH₈C₂H₅ CICH₂OC₃H₆H₈OC(O)CH₃

Slimicide for paper mills Seed treatment, turf, soil Seed treatment Seed treatment, slimicide, soil disinfectant, paint preservative

TABLE 1-3 **ORGANOMERCURY FUNGICIDES**

Phenyl mercurials

Phenyl mercuric acetate

Tolyl mercuric acetate Phenyl mercuritriethanolammonium lactate Phenyl mercurimonoethanolammonium acetate Phenyl mercury urea

Phenyl mercuric dimethyldithiocarbamate

C₆H₅HgOCOCH₃

CH₃C₆H₄HgOCOCH₃ C₈H₅HgN(C₂H₄OH)₃·OC(O)CH(OH)CH₃ C₆H₅HgNH₂C₂H₄OH·OC(O)CH₃ C₆H₅HgNHC(O)NH₂ C₆H₅HgSC(S)N(CH₃)₂

Seed, turf, foliage, and industrial fungicide

Seed, bulb, turf, foliage treatment

Scab treatment on fruits, ornamentals, turf Scab treatment on fruits, ornamentals Seed treatment Slimicide for paper mills and mold retardant for paper

Phenoxy mercurials

Hydroxy mercurichlorophenol Hydroxy mercurinitrophenol

Methyl mercury 8-hydroxyquinolate

CIC₆H₄OHgOH $O_2NC_6H_4OHgOH$

ŎHgCH,

Ethyl mercurithiosalicylic acid .COONa

SHgC2H3

Alkyl mercurials

Methox yethylmercuric chloride Ethylmercuric chloride

CH₃OCH₂CH₂HgCl C₂H₅HgCl

Seed and bulb fungicide

Disinfectant for potatoes

Seed fungicide

Seed treatment Seed treatment

Ethylmercuric acetate 1-Acetoxymercuri-2-hydroxyethane Methylmercuric dicyandiamide Ethylmercury 2,3-dihydroxypropyl mercaptide Chloromethoxypropylmercuric acetate

Miscellaneous mercurials

N-(Ethylmercuri)-p-toluenesulfonamide

N-(Ethylmercuri)-1,4,5,6,7,7-hexachlorobicyclo-[2.2.1]-hept-5-ene-2,3-dicarboximide (also mercurymethyl derivative)

C2H5HgOC(O)CH3 HOC₂H₄HgOC(O)CH₃ CH3HgNHC(=NH)NHCN HOCH2CH(OH)CH2SHgC2H5 CICH2OC3H6HgOC(O)CH3

CH₃C₆H₄SO₂N(C₆H₅)HgC₂H₅

Slimicide for paper mills Seed treatment, turf, soil Seed treatment Seed treatment, slimicide, soil disinfectant, paint preservative

Seed treatment

Seed treatment, lawn

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organic group R appears to regulate the transport and stability of the compound, and the anionic group determines the solubility. Thus, Booer (10) has described ethyl- and methoxyethylmercury radicals as resembling sodium ion in that they are strongly alkaline and form highly ionized salts that are generally water soluble and appreciably volatile. These compounds are quantitatively decomposed by strong acids:

$$CH_3OCH_2CH_2Hg\cdot CI + HCI \longrightarrow C_2H_4 + CH_3OH + HgCl_9$$

In contrast, the phenyl- and tolyl-mercury radicals resemble silver in giving very stable and generally insoluble salts that can be melted and sublimed or boiled with strong acid without decomposition.

In soil all the organomercury compounds are decomposed to mercury salts or to metallic mercury which are the active fungicides. Booer(10) has suggested that this decomposition takes place through base exchange to form organomercury clays which subsequently form mercury salts by further base exchange. These mercuric salts are then reduced to mercurous salts and to mercury. Typical reactions suggested are:

(1)
$$(C_2H_5Hg)_2$$
— $Clay \longrightarrow (C_2H_5)_2Hg + Hg$ = $Clay$
 Hg = $Clay + CaO \longrightarrow Ca$ = $Clay + HgO$

O
H

(2) $2CH_3OCH_2CH_2HgOCCH_3 + Clay \longrightarrow (CH_3OCH_2CH_2Hg)_2$ = $Clay + HgOCCH_3COCH_3$

$$(CH_3OCH_2CH_2Hg)_2$$
— $Clay + 2CH_3COOH \longrightarrow Hg$ — $Clay + 2CH_3OH +$

$$O \\ \parallel \\ 2C_2H_4 + Hg(OCCH_3)_2$$

The metallic mercury liberated in the soil is ultimately converted to mercury sulfide by reaction with H₂S liberated by soil microorganisms.

2. Structure-Activity Relationships. As we have seen, the organomer-cury fungicides can be represented by the general formula RHg·X and the organic radical and anionic group determines the stability, transport, and solubility of the compound. Perhaps the most thoroughly studied series of compounds is the alkoxyalkylmercury salts (89). The methoxy-and ethoxyethyl-mercury compounds were of equal effectiveness as fungicides and had the most favorable ratio between "curative dosage for fungi" vs. "tolerated dosage for seed germination." The propoxy-,

ment.

stability of the y. Thus, Booer als as resembhighly ionized le. These com-

mble silver in be melted and

led to mercury les. Booer(10) base exchange ercury salts by ced to mercur-

Clay +

2CH₃COOH ЭH+

/ converted to organisms.

he organomerla RHg·X and lity, transport, oughly studied The methoxyfectiveness as irative dosage The propoxy-, isopropoxy-, butoxy-, and isobutoxyethyl mercury compounds were also effective fungicides but had less favorable curative vs. tolerated ratios. For the anionic portion of the molecule there was no general difference in effectiveness as dry seed treatments between inorganic salts and salts of carboxylic acids, which were effective at usual dosages in preparations containing 1.5% mercury. When the free valence of mercury is attached to moieties containing N (CH₃OCH₂CH₂HgN-) or S (CH₃OCH₂CH₂HgS-), activity was decreased so that 2% mercury was required in dry seed treatments. If the free valence of mercury is attached to a second carbon as in CH₃OCH₂CH₂HgC=CHgCH₂CH₂OCH₃, activity is still further decreased so that a content of 3% mercury is required for dry seed treat-

- 3. Mode of Action. The mechanism of toxicity of mercury is not very specific and the mercurial fungicides all owe their activity to the Hg²⁺ moiety. In the organic mercurials the alkyl or aryl portion serves to conduct the Hg²⁺ to the site of action, by reason of lipoid solubility, and also determines the stability and rate of release of the mercury ion. The biochemical action of the mercury fungicides is related to the affinity of Hg²⁺ for the sulfhydryl groups of essential respiratory enzymes. Thus mercury treatment decreases the oxygen uptake of treated fungus spores, and poisoned spores can be revived by subsequent application of sulfhydryl compounds such as glutathione or cysteine. The specific mercury chelating agent BAL or 2,3-dimercapto-1-propanol is also effective in protecting fungus spores against mercury (91).
- 4. Biological Activity. The mercury fungicides (see Table 1-3) are generally applied as seed dressings used either as dusts or slurries containing from 1.5 to 3.2% metallic mercury for application to seeds of cotton, rice, wheat, flax, peanuts, safflower, and other crops. They are also used as foliar fungicides for seab of apples pears, strawberries, and other fruits; for the treatment of gladiolus corms and other bulbs; and as protectants for potato seed pieces; and for control of fungus diseases of turf. Certain of the compounds are very effective slimicides for paper mills, and are mold resistant treatments for paper and paints.
- 5. Toxicology. The organomercury compounds are of high to moderate acute toxicity to animals with oral rat LD_{50} values of ethylmercuric chloride 30, phenylmercuri triethanolamine lactate 30, methylmercury 8-hydroxyquinolate 72, ethylmercurithiosalicylic acid 75, N-(ethylmercuri)-1,4,5,6,7,7-hexachlorobicyclo-[2.2.1]-hept-5-ene-2,3-dicarboximide 148, methoxyethylmercuric chloride 570, methoxyethylmercuric silicate 1140, and phenyl mercuric acetate 2080 mg/kg. These values compare with the oral LD_{50} of 37 mg/kg for mercuric chloride.

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3. Me tin com hibitors the inco place in

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aryl tins, are among the most effective fungicidal agents yet discovered. These organo-tin compounds are very general biocides and are toxic to plants, insects, and various marine organisms. At present tributyl tin hydroxide is used as an anti-mildew agent in wood, textiles, and paints: in antifouling paints for marine vessels; for prevention of microbial slime in paper mills; and as a germicide. Triphenyl tin chloride, m.p. 105°C, is used in antifouling paints; and triphenyl tin hydroxide, m.p. 120°C,

> (C₄H₉)₃SnOH (C₆H₅)₃SnOCCH₅ tributyl tin hydroxide triphenyl tin acetate

(C₆H₅)₃SnCl

(C₆H₅)₃SnOH triphenyl tin hydroxide

triphenyl tin chloride

Tin is a group IV element and forms a limited series of organic compounds analogous to those of carbon. In the stable compounds the tin is

From chronic exposure the mercury compounds are highly toxic and

diets of phenyl mercuric acetate as low as 0.5 ppm produced kidney

damage in animals. The nature of the organic radical determines to some

extent the amount of tissue storage and damage in the liver and kidney:

inorganic mercury < aryl mercury < alkyl mercury. The concentration

of all three classes increased after daily oral dosage in the following

order: brain < blood < liver < kidney. In man, long exposure to organic

mercurials causes nervous symptoms: tremors, incoordination, disturb-

Mercury compounds can enter the body through skin absorption, by

mouth, and by the respiratory tract. The appreciable volatility of some of

the alkyl mercury compounds increases the hazard of inhalation, and the

cause severe dermatitis. Protective clothing should be used when hand-

ling these materials and scrupulous personal hygiene is required. Mercury

compounds pose a severe environmental hazard through concentration

and accumulation in food chains as biologically formed dimethyl mercury.

1. Introduction and Chemistry. Investigation of the fungicidal activi-

ties of tin compounds was begun in 1950 by Van der Kerk and co-workers

(146). It was soon found that although inorganic tin compounds were in-

active, certain organo-tin compounds, particularly the trialkyl and tri-

and triphenyl tin acetate, m.p. 121°C, are used as agricultural fungicides.

The organic mercury compounds are severe skin irritants and may

ance of hearing, irritability, and severe mental disturbance (55).

threshold limit for organic compounds in air is 0.01 mg/m³.

H. Organo-Tin Compounds

MEMO

SUBJECT ____MONROE TWP MUNICIPAL WELL #4 & #5

NEW JERSEY STATE DEPARTMENT OF ENVIRONMENTAL PROTECTION

го	FILE								
FROM	FRANK FA	ARANCA, F	HSMS IV	, BUREAU	OF	PLANNING	AND	AS SAESSMENT	

On 4-9-87 and 4-13-87 the writer conducted a site reconnaissance of Monroe Township Municipal Wells #4 & #5.

The writer met with Mr. George Cassabone, Supervisor of Monroe Twp. Municipal Utilities Authority on 4-9-87. Mr. Cassabone and the writer conducted a windshield survey of Wells #4 & #5, of which #4 is not currently in use. He explained that Monroe Township M.U.A. currently has 4670 services and the current population is 26,000 people.

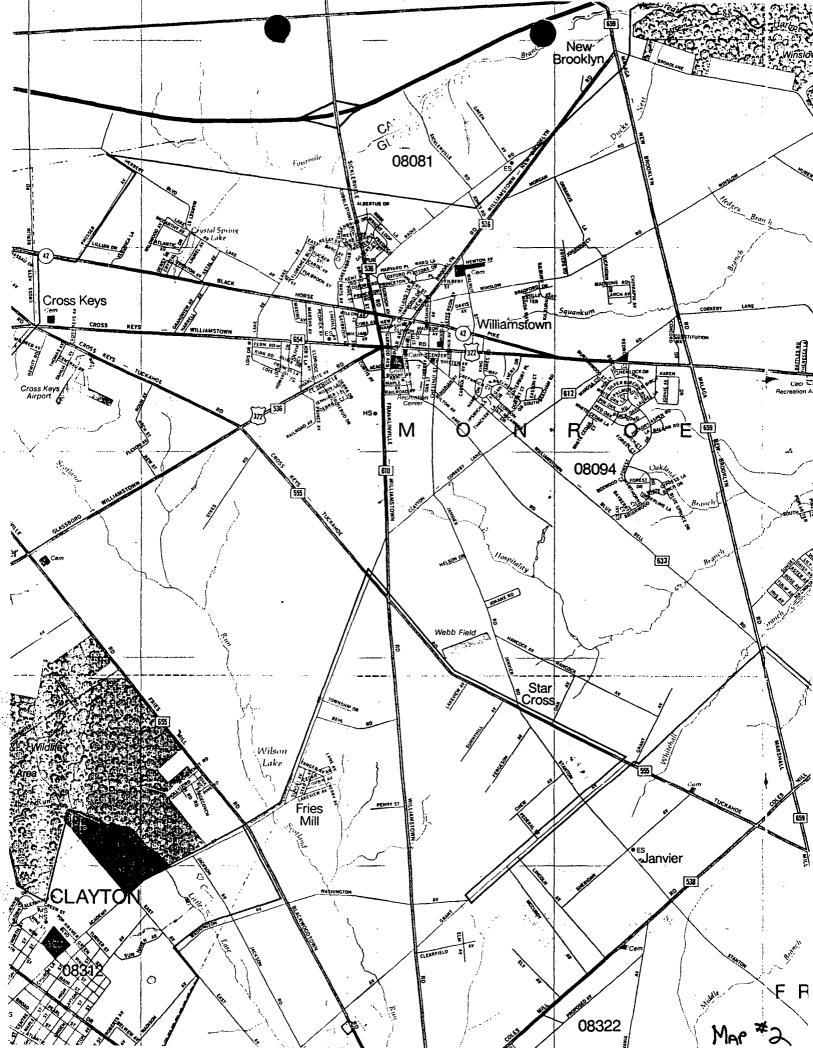
On 4-13-87 the writer investigated several areas within Williamstown as possible sources contributing to the present mercury contamination of the ground water. Two sites are believed to be suspect for further investigation.

- 1. The Atlantic City Electric Company Williamstown Substation
 This parcel of land is located adjacent to Clayton Road (Franklinville
 Williamstown Road) on Block 13301, Lot 14. The owner is located at
 P.O. Box 1500, Pleasantville, NJ 08232. The site is 2.06 acres and
 contains several rows of oil circuit breakers, transformers and a
 capacitor bank. The soil within the substation is void of all
 vegetation and appears stained in several areas.
- The Gravel Pit This parcel of land is also located adjacent to Clayton Road on Block 13301, Lot 11 & 12. The owner is Thomas H. Webb Jr., Rd 7, Box 175, Jackson Road, Williamstown. This site is approximately 10.24 acres which is an inactive gravel pit. The pit itself has had unauthorized dumping of assorted solid waste over many years. A large portion of the gravel pit is devoid of vegetation, and is currently used by the neighbors for the use of recreational all terrain vehicles. The solid waste is mostly old concrete, asphalt, soil, landscape debris and household trash.

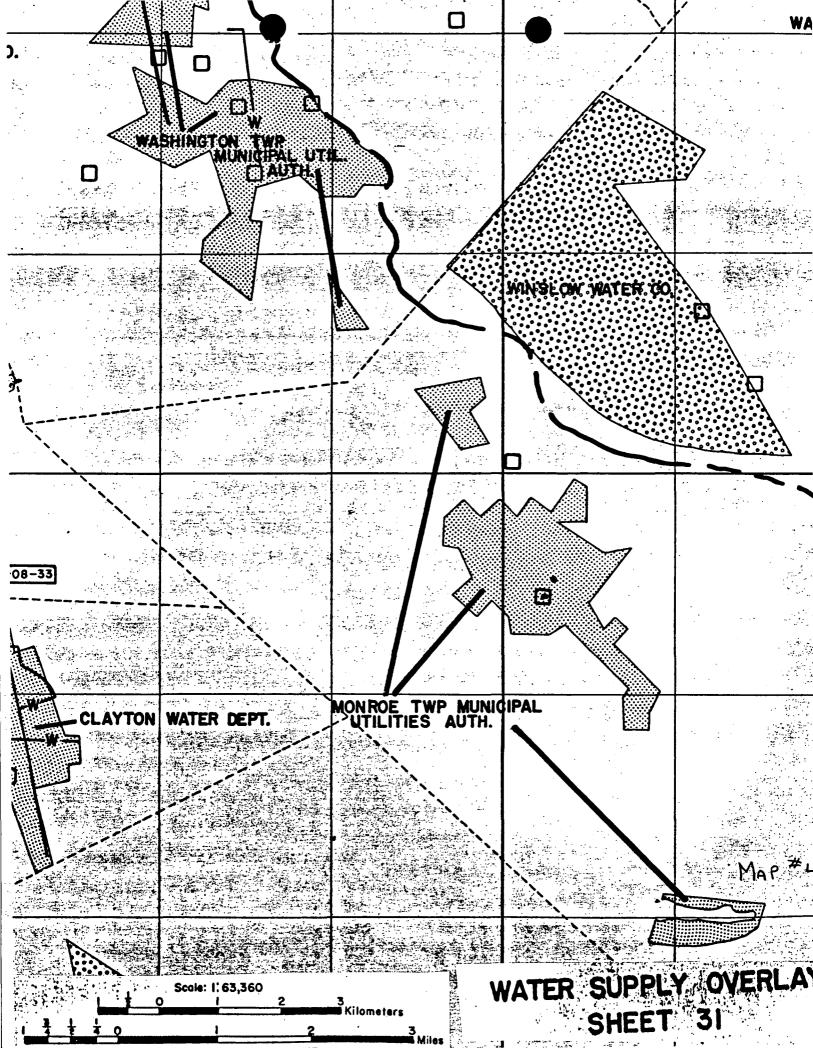
These two sites were in operation prior to the detection of mercury in Well #4 in 1976. An analysis of the USGS Topographic Series Maps reveal that these sites are prime canidates as sources of contamination based upon the hydraulic movement of surface water and ground water to down gradient Wells #4 & #5.

Also suspect is the land on which the Williamstown High School is now located. In the past Mr. Frank Simei owned the land and used it as a farm for spinach and lettuce. However, the use of mercury as a fungicide is not usually associated with these two food crops.

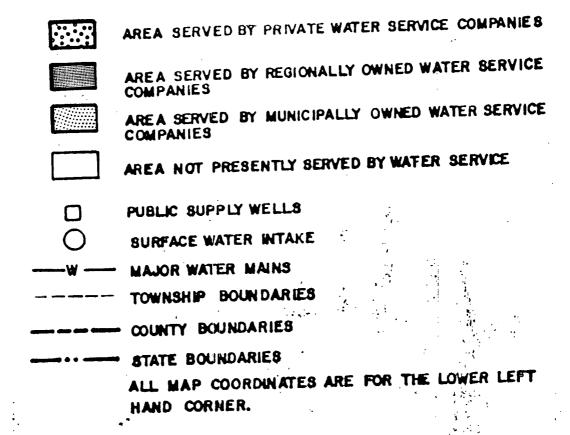
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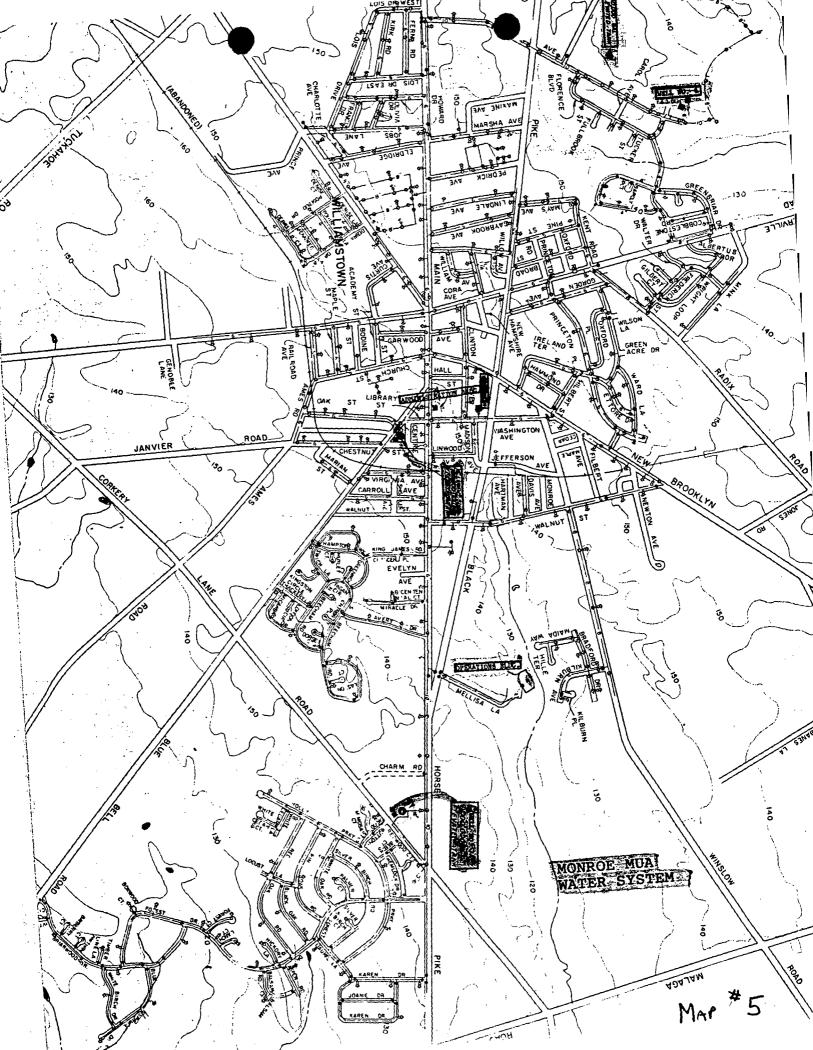


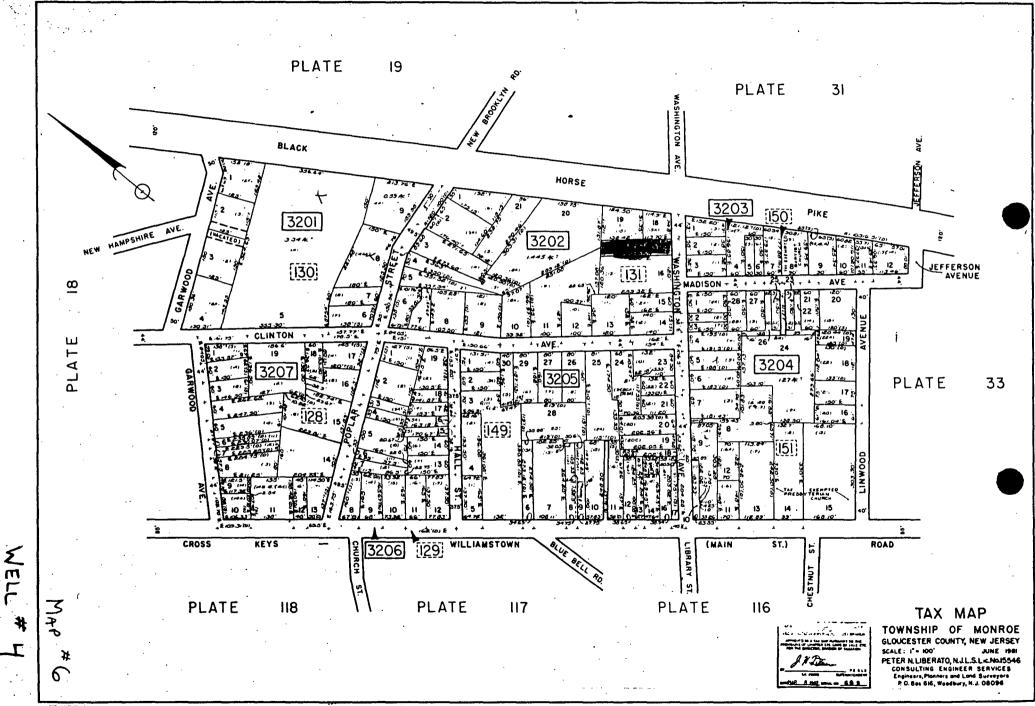
LEGEND



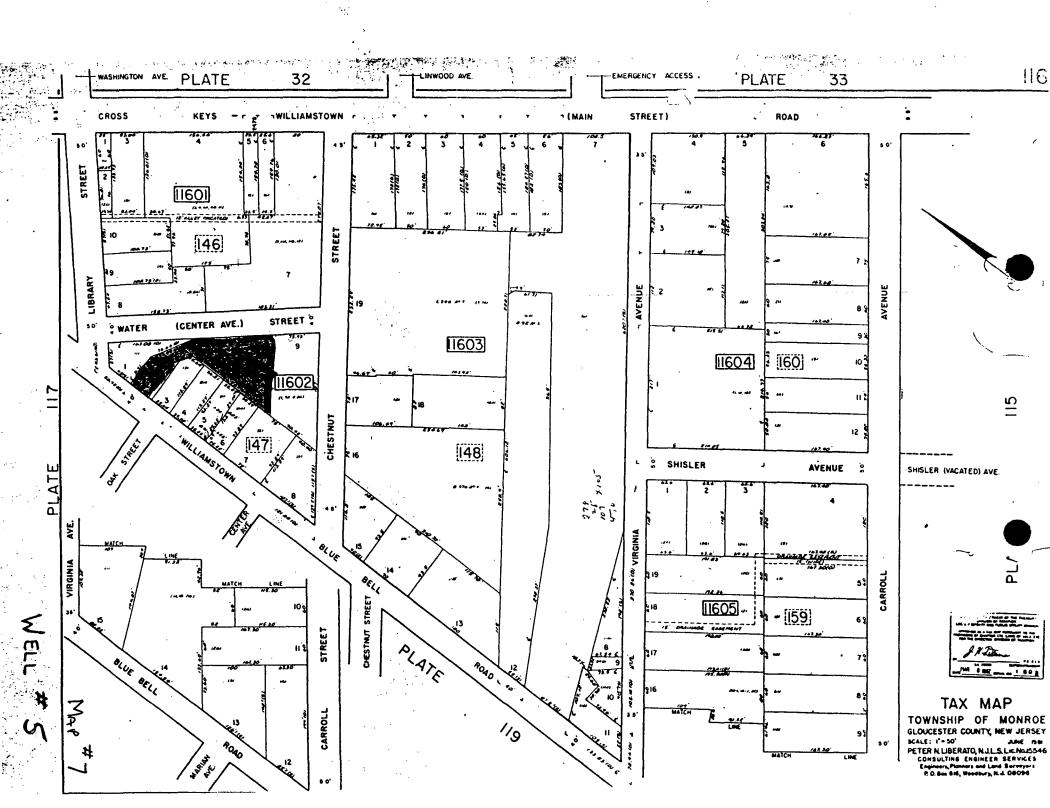
LOCATION AND OWNERS OF PUBLIC SUPPLY WELLS

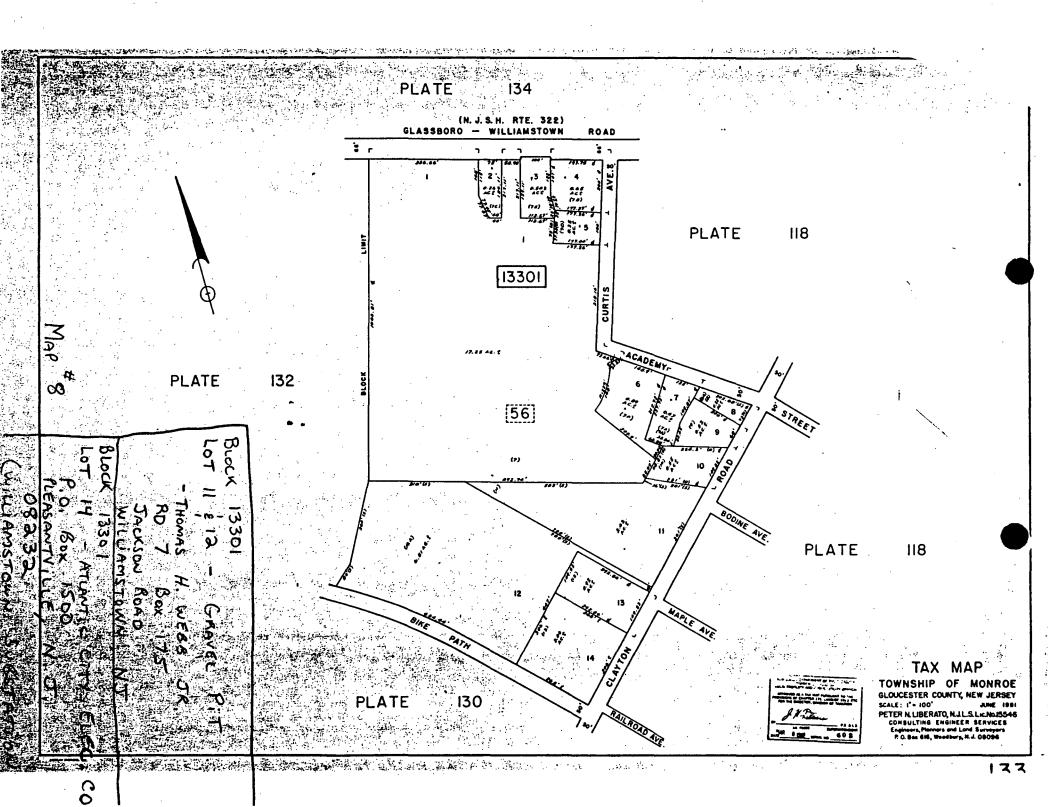
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Borough of Berlin
31-23-236
31-23-236 Borough of Berlin
31-23-344
          Borough of Berlin
31-23-347
          Overbrook High School
          Lower Camden Regional High School
31-23-367
          Ivystone Water Works
31-23-395
          Monroe Township Municipal Util. Auth.
31-23-777
          Winslow Water Co.
31-23-818
          Winslow Water Co.
31-23-858
31-23-899 USGS, New Brooklyn
          Assumption Parish School
31-24-158
31-24-177
          Ivystone Water Works
          Camden Co. Board of Education
31-24-792
          Borough of Glassboro
31-31-322
31-32-174 Borough of Clayton
31-32-441
           Borough of Clayton
          Monroe Township Municipal Util. Auth.
31-33-146
31-33-954
          Wharton Realty Co.
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Form 87-54-4-49

DEPARTMENT OF CONSERVATION AND ECONOMIC DEVELOPMENT Division of Water Policy & Supply WELL RECORD

Permit No. 31-361
Application No. 7.29.
County Glowsester

7/ 22 126 1

1. OWNER Monroe Township ADDRESSVIlliamstown, Gloucester County, NJ Owner's Well No. 4 SURFACE ELEVATION Feet
(Above mean sea level) 2. LOCATION Washington Ave. Pumping Station 8. DATE COMPLETED 11/12/51 DRILLER A. C. Schultes & Sons Woodbury N.J 4. DIAMETER: Top. 12" Inches Bottom 12" Inches TOTAL DEPTH 106 Feet 5. CASING: Type Blk Steel Diameter 12" Inches Length 67' 11" Feet Size of .030 6. SCREEN: Type Johnson Opening .. 0.60. Diameter .. 12" ... Inches Length .41! 9" Feet Range in Depth Top 69 Feet Geologic Formation Cohansey. Length Feet 7. WELL FLOWS NATURALLY No Gallons per Minute at Feet above surface Pumping level. 321.5". feet below surface after 24 hours pumping Drawdown 20 Feet Specific Capacity 40. 25 Gals. per min. per ft. of drawdown How Pumped Turbing Test Fump How measured Orifice 9. PERMANENT PUMPING EQUIPMENT: Horse Power R.P.M. How Driven Feet (Average..... Gallons Daily 10. USED FOR Public Supply AMOUNT Maximum Gallons Daily Taste.....None... Odor None Color..clear Temperature. 58.... 12. LOG See Reverige and an lest of that of an approximately Are semples available?

18. SOURCE OF DATA ... Driller's.Log

14. DATA OBTAINED BY A. C. Schultes Jr. DATE 11/15/51

Log of Well.

```
F111
                 Brown Sand
            61
                Brown Sand & Stones
           12'
  12'
           17
                 Gravel & Stones_
  171
           281
                Coarse Sand & Gravel
                Fine Sand & Clay
  281
           33'
 331
           351
                Coarse Yellow Sand
 351
           441
                Yellow Clay
Coarse Sand
           561
 561
          581
                Fine Yellow Sand
 581
          661
                Yellow Sand
 661
                Yellow Sand
Yellow Sand
          71'
 71'
          741
 741
          761
                Yellow Sand
 761
               Sand & Gravel
Dark Yellow Sand & Gravel
          81!
 81'
          961
               Dark Yellow Sand
 96'
         104'
104'
         106'
              DarkBrown Sand cemented
106' -
         107' Black Muddy Clay
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RECEIVED NOV 9 951

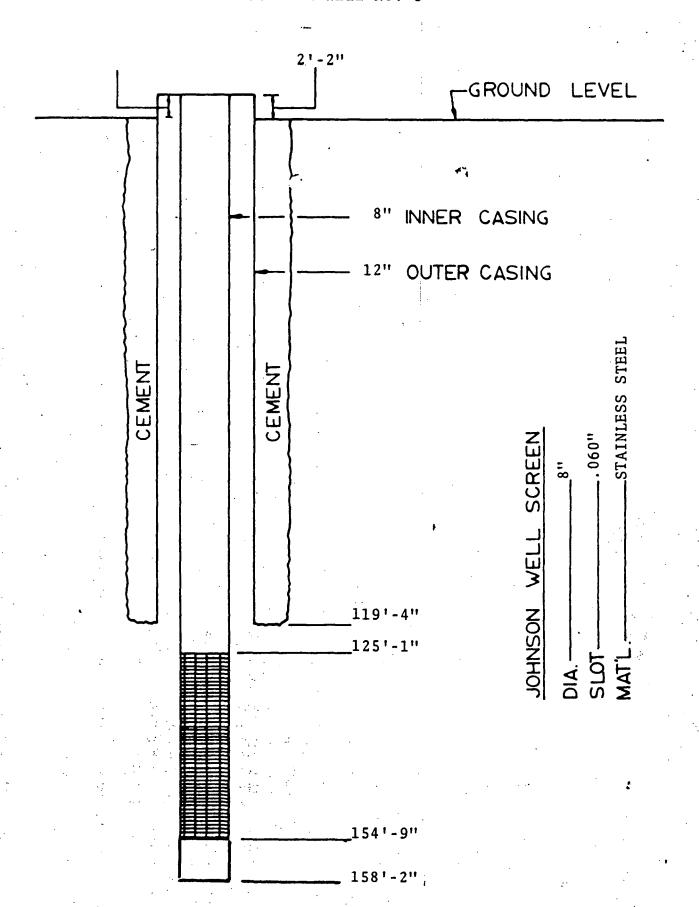
A. C. SCHULTES & SONS

Water Well Contractors

Gravel Packed Well

;	· .	Ordiver Little	GEL CAGIRON	·
1 . 1	11			
1-1	Ī	WELL LOG	FELT FROM GOODND	MONROE TOWNSHIP MONIC
SPOC 10	ievic	Pill	0 to 1'	PAL UTILITIES AUTHORIT
. 1 1 1	1	Sand (laminated)	4 - 8'	Well No. 5
		Sand & stones	8 - 51'	Jeb No 7194
		Sand (some laminated)	51 - 58 +	Location Williamstown, M.I
		Clay (black)	58 - 63'	Test Puniped (Hrs.) .9% hrs.
-	, i	Sand	63 - 68'	Capacity G.P.M
•	2	Clay (black)	68 - 791	Static Level (Ground) 36 2"
		Sand	79 - 84'	Pumping Level (Ground: 82)
		Hardpan	84 - 851	Specific Capacity . • 11
		Sand	85 - 87'	Diagreter of Outer Casing 12"
		Clay (laminated)	87 - 92'	Diameter of 8 II
	. 1	Sand/thin layers clay	92 - 116'	Dept of Well (Ground 160
•		Sand/thin layers hardpa	n 116 - 126'	Dept to R. L. Nipple (Ground)
7 1 A		liardpan	126 - 127'	Depth to Gravel (Ground) 201
		Very hard packed sand	127 - 155'	Grave Size : #3
- L	<u> </u>	Fine sand	155 - 157'	Length of Chains 12" = 121"
1.2		Clay	157 - 180'	Length of Casing 8" = 126"
				Underresm Size 32
				Type : Screen Johnson S/S
		,		Size of Screen (Div.) 8" I.D.
i.				Top Streen Fitting Coupling
1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1				Botton Screen Fitting Plug
Tark Sand		7 72	The second dispussion of the second s	51gt 5 :- #60
				Brank No
				Big + Sement 100
				Drotte : Machine 1250
1		\ \ \ \ \ \ \ \ \ \ \ \ \ \ \ \ \ \ \		Date Well Completed 1/18/66
4 3	1			Doller Hammond
1 1 1				100 100 100 100 100 100 100 100 100 100
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MONROE TOWNSHIP MUNICIPAL UTILITIES AUTHORITY WILLIAMSTOWN, NEW JERSEY JOB NO. 7194 PRODUCTION WELL NO. 5



A.C. SCHULTES & SONS, INC.

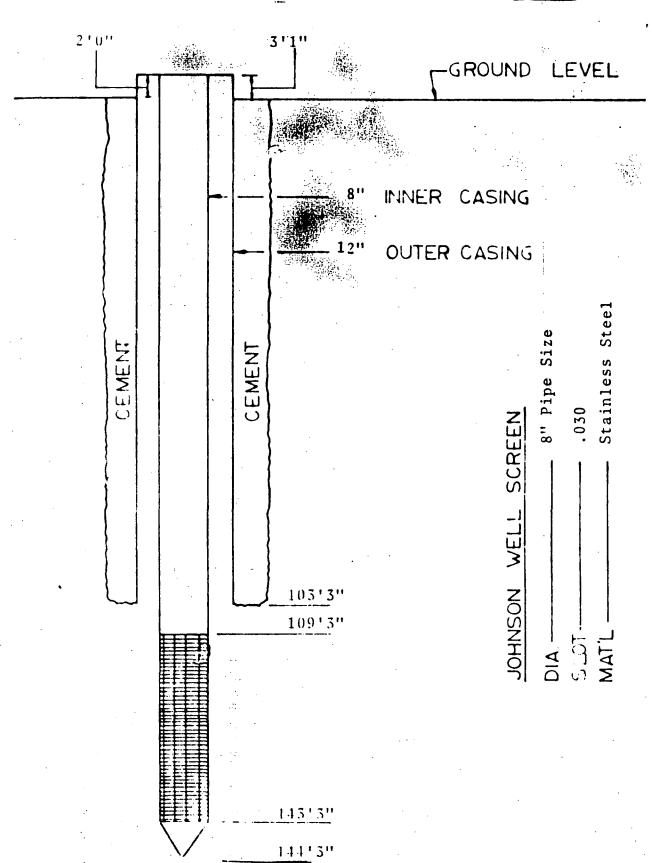
MONAGE TWP.
MUA #6

GRAVEL PACKED WELL

A * 6	,	GRAVEL PACKED W	L. L.	
	 	Top soil-	FEET FROM GROUND SURFACE	NAME OF OWNER Freeways Properties
GROUND	LEVEL 1	Lt. yellow sand &	2'-4'	Monroe Woods Location Williamstown, !
				# 1
		stones	4.	
		Dr. yellow sand	4'-17'	Job No. 7896
	i.	clayish		Test Pumped (Hrs.) 8
		Yellow sand &	17'-22'	Capacity G.P.M. 412
		stones clayish		Static Level (Rotary Table) 15 10"
↑		White sand & stones	22'-36'	Pumping Level (Ratary Table) 41 1
		med. to coarse		Specific Capacity 10
		Clay, lt. brown	36'-42'	Diameter of Outer Casing 12**
	1 -	Sand	42'-51'	Diameter of Inner Casing 811
	,	Hard clay	51'-53'	Depth of Well (Rotary Table) 146 ' 3"
H d		Sand & gravel	531-801	Depth to R.L. Nipple (Rotary Table)
101 4 1-101 A 101		thin layers clay		Gravel Size #3
2	· -	Hard packed sand	80'-101'	Length of Outer Casing 106 '4"
	 -	ξ gravel		Trength of Inner Casing and Screen 145 '3"
	·- ·	Red clay	101'-102'	Ginderream Size 31*1
		Sand & gravel	102!-127!	Screen Material S.S.
		Sand, fine med. gr.	127'-141'	Johnson Johnson
	-	Fine green clayish	141'-160'	. Telescope Size of Screen (Clus): Y Pipe Size (
		sand		Length of Screen 34 1
	· · · · · · · · · · · · · · · · · · ·	Clay greenish ! tack	160'-166'	Top Screen F.I.P.T.
	_			Softom Screen M.I.P.T.
		**1		Siot Size .030
January W.	1.			Bigs of Cement 105
14 6 m	<u> </u>			Drilling Machine 1250
	· . \.			Completed 1/30/70
The state of the s				Doller Albert Hammond
•	`			

FREEWAYS PROPERTIES MONROE WOODS MONROE TOWNSHIP WELL # 1





Organi

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Application No.

WELL RECORD

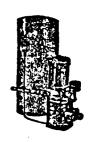
OWNER Violet	Packing Compan	ADDRE	ssi23 Railre	ed Ave. Will	iametown
Owner's Well No	. 31-5150	SURFA	CE ELEVATION	150 Above even see 10	Eget
	ailroad Ave. W				
DATE COMPLETED _	8/15/67	DRY LER	J. G. Holma	nice training of the control of the	
DIAMETER: top_	8Inches Bot	tomInche	TOTAL	DEPTH 143	FORE
	lack iron	and the second s	Control of the Contro		· · · · · · · · · · · · · · · · · · ·
Market State Comments	inless steppining				
		· 一日一日本部一門東北海山大田東海	and Artifaction of the control of th	7.00	4. 3.12
Andse in Depth	Top 115 8ottom 150	640logic	Formation		
	amotor	And a section of the			
11.4	RALLY Gallo		See Addition to the control of the second		
Mater rises to_		Feet above so		Leef above	BUT THE CRA
4 6 4 4 A C	Dato 6/13/18	ranger and the first terms of the second			
4	evel before pumpin	on Market & Commercial			
an géalla de Prake 1996 — la più la géalla de Paris de P		the second secon	. :	Feet below	
	Feet Spe				
	Turbine 300	A State Control of the Control of th			7 7 7
ROW Punped	t on nearby wells.		ow measured	, and	
	•	× 1		Signal Landschaff (1985)	
PERMANENT PUMP	sible arrice				-
	3CO 6.P.M.	_ MITS. NAME	ectricity.	20	3500
the second					.W Feet
Bepth of Pum.	in well to	eet Type of	rootylece in wa	111 <u></u>	F • • 1
Wepth of Alf	t ne in wellF	eet lype of		•	einches
WSED FORIr	nustrial coolir	AMO	UNT { Average	GAIIO	ns Daily
TATE OF THE OF THE	ATER GOOG		(Maximum	Gallo	on a U a l'Try
QUALITY OF W	AIEKGOOG		Sample: Y	es No	
Jests	Qdor	Col	Are sai	Temp	2
- MILUU	The Little And Little State St				

OBTAINED BY



SOUTH JERSEY DRILLING CORP.

South Delsea Drive Vineland, New Jersey 609-692-7854



Violet Packing Company Williamstown, New Jersey

TOG

0 - 1	Topsoil
1 - 3'	Yellow clay & large gravel
3 - 11*	Sandy clay
11 - 13*	Clay
13 - 21.	Fine to medium yellow sand
21 - 28*	Coarse sand w/fine gravel
28 - 30*	Clay
30 - 36"	Medium to coarse sand
36 - 43*	Alternating thin layers of clay and sand
43 - 51"	Pine to medium sand
51 - 60'	Yellow & gray clay
60 - 64'	Medium sand w/some gravel
64 - 691	Gray clay
69 - 87*	Sand & gravel
87 - 115	Black clay
115 - 150*	Coarse sand & fine gravel

135

WELLS

NIDUCTOU

DALIESTI.

016171611 811118

Municipality Collected by a to	TO THE PARTY OF TH			Memo # W-70					
BACTERIOLOGIC	AL ANALYSES Coli	iform organisms, orine residuals a	determined by t	he membrane filter om.	technique are	reported in co	olonies per	100 ml	
Sample Number		Point of (Collection			Organisms 💝		Residua Total	
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611	t /p//# 9	4.4.4	e bille				See To To A	~~	
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			be Vaer San	cept color odor t arcs		n. Figures in	parentries	s are in	
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Turbidity					C.	dmlum		net-	
**					i iz≟ c	romium +6		Mr	
Alkalinity to pH 4					07.45.25.25	opper *	A Contraction	0.14	
Nipetras NO3 (30	NEW TRANSPORT AND THE PARTY OF				CHANGE SY	/anide	t- 69.2554	ALT-	
Chloride (250 Total Cir Solids (500	The state of the s					ercury	Section 1	7,70 Z 2.0 pp	
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Manganess (0.0								16	
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Sulphate 26 4 250				4		GLOUCE	STER	CO.	
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						5 C 3 X			
ionalia 2									
TOTAL STATE OF THE			A SAN AND A SAN					4 mg = 1	

DEPARTMENT OF ENVIRONMENTAL PROTECTION	6-22-16
County	Couralle
Coality Date Collect	ed 6/21/76
ected/by Wemo # U = 7 E Project # W-	
TERIOLOGICAL ANALYSES: / Coliform organisms determined by the membrane filter technique are reported in control of the collection of the c	olonies per 100 ml.
Coliform Organisms	Chlorine Residual

E. W. C. W. LOND TO STATE OF S	Merchanism and the second	andas agree as a green of the green				the second section of the second		· · · · · · · · · · · · · · · · · · ·		
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YSICAL:—CHEMICAL ANALYSES:—Determinations are in ppm except color odor, turbidity, and pH: Figures in parentheses are from the N: I. Potable Water Standards

				一种人工工工工工工工工工工工工工工工工工工工工工工工工工工工工工工工工工工工工			
mple Number≱ 😪	17981				27.04	Sample Number	17981
olor (i)	306,2					Arsenic	J. Million
LOT SECRET STORY		學師為為				Barium	/ NEG-
rbidity [5]				Person		Cadmium	1 ME
产品组成本	5 51			建筑	治疗与德	Chromium +6	1 Mon
ikalinity to pH 4	9	#37-12-13	All Control			Copper	10.063
itrate as NO3 * (30)	Yeay.		***			Cyanide	MEi-
hloride (250).				X		Lead	0.006
otal Dis Solids (500).	The second of the second of the second			M		Mercury	1.8pp[
BS/LAS (0.5)	Commence and transport					Selenium	Mo
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SHEET 2 - WATER PURVEYOR

Department of Environmental Protection MEMO Division of Water Resources Mr. Richard Bellis, Assistant Director, Monitoring, Surveillance and Enforcement Element Mr. John Wilford, Assistant Director, Water Supply and Flood Plain Management Element DATE November 3, 1976 SUBJECT Mercury Contamination of Ground Water - Monroe Township, Gloucester County

State of New Jersey

Attached is a copy of a memorandum-report which discloses mercury contamination of the water derived from Wells #4 and #5 of the Monroe Township Municipal Utilities Authority, Gl ucester County. As is noted therein, based upon an initial, cursory investigation, personnel of the Bureau of Potable Water have been unable to determine the origin of the mercury. No manometers, or other instruments utilizing mercury, are in use at either of the wells.

I have advised the M.U.A. to immediately undertake the construction of a new well or wells to replace Wells #4 and #5 and, in the interim, to use Well #6 as the main production well, augmented as necessary by Well #5 to meet daily water demands. I have suggested that Well #4 be taken out of service and be retained only for standby purposes to meet a dire emergency such as a major fire.

On the basis of a specific test made on October 5, 1976, the mercury is in the inorganic form. This, fortunately, has less toxic potential than the alkyl (organic) form. However, the National Interim Primary Drinking Water Regulations promulgated by EPA pursuant to the Safe Drinking Water Act, which will become effective in June 1977, impose a maximum contaminant level for total mercury (0.002 mg/l), and make no differentiation between the inorganic and organic forms.

On the basis of the findings by the Bureau of Potable Water, it is requested that you will refer the matter to your Office of Special Services and have them conduct an intensive survey to determine, if possible, the origin of the mercury. Dave MASAN

From h MASAN

JW:bn

Enclosure

cc: Mr. Zelikson

Mr Bellic

Mr. Wilford

Messrs. Laffey and Vora

November 3, 1976

Mercury Contamination, Monroe Township Municipal Utilities Authority

The recently instituted program for sampling raw water sources throughout the State has disclosed a mercury contamination in two wells owned and operated by the Monroe Township Minicipal Utilities Authority, Gloucester County, New Jersey.

The three operating wells of this system were sampled on June 21, 1976. Wells #4 and #5 showed mercury values of 2.0 ppb and 1.8 ppb respectively, while Well #6 yielded a negative result. The mercury contamination in Wells #4 and #5 was confirmed by samples taken on July 19, August 18, August 24, and October 18, 1976 with results ranging from 1.4 ppb to 6.4 ppb in Well #4 and from 0.8 to 2.8 ppb in Well #5. On October 5 and October 15, 1976 the water from Well #5 was negative for mercury, but Well #4 showed values ranging between 1.4 ppb and 2 ppb. The October 5 sample from Well #4 showed that the mercury was all of the inorganic tupe. System samples taken on August 18, August 24, October 10 and October 18 showed mercury values ranging from 1.0 ppb to 10.8 ppb. A tabulation of the analytical results is attached.

An inspection of the Monroe Township M.U.A. system was conducted on October 5, but no determination could be made for an immediate source of the mercury. Walls Nos. 4 and 5 are located respectively in Washington Avenue and Chestnut Streets, with an intervening distance of 1,200 feet.

Well #4 is the main production well for the system, having a reported yield of 800 gpm. It was constructed in 1952, taps the Cohansey Aquifer, and is cased to its full depth of 106 feet. Our records do not indicate whether or not the annular space is sealed. The formation log shows a clay layer between 35 feet and 44 feet.

Well #5 was constructed in 1967 and has a reported yield of 500 gpm. It taps the Bridgeton Tertiary Cohansey aquifer, and is 160 feet deep. The annular space between the casing and the drill hole is sealed to a depth of 127 feet. The formation log shows the presence of a clay layer between 58 feet and 63 feet, and a layer of hardpan between 84 feet and 89 feet.

A tour of the area surrounding Wells Nos. 4 and 5 failed to indicate any probable sources of mercury contamination with the exception of the Violet Packing Company located 0.5 miles west of Well #5. This company is engaged in the production of tomato sauces. It operates seasonally during the summer months and, during operations, produces about 200,000 gallons of wastewater

per day which is treated by pH adjustment, aeration and settling, prior to disposal either to the sanitary sewer or on to adjacent farmland. During the 1976 season their wastes were disposed of to the ground.

There are several sewing factories located in various parts of the Township, engaged in the manufacture of clothing, but these are believed to be all dry industries. There is a sanitary landfill located approximately one mile north of wells #4 and #5, but this is not known to receive chemical wastes.

A total of 24 public water supply wells in the area surrounding Monroe
Township was sampled, all with negative results for mercury. A tabulation
of these wells is also attached.

The results of the various samples taken show, conclusively, that the water from Wells #4 and #5 contains mercury, and that this constituent is also present in the delivered water. Current New Jersey Potable Water Standards and the 1962 P.H.S. Drinking Water Standards do not include a maximum contaminant level for mercury, though there is a "rule of thumb" maximum of 2.0 ppb. The recently-promulgated National Interim Primary Drinking Water Regulations, however, which will become effective in June 1977, include a maximum contaminant level of 2.0 ppb, for mercury, and though inorganic mercury is considered to be of far less toxic potential than organic mercury, the imposed value is for total mercury.

The operator of the system has been informed of these findings but, to date, they have not been officially brought to the attention of the Monroe Township M.U.A. It is, therefore, our recommendation that they be apprised of the situation so they can take immediate steps to develop alternate sources of water prior to the effective date of the National Interim Primary Regulations and thus avoid the appropriate public reporting requirements and the necessity for applying for an exemption or waiver in accordance with the requirements of the federal Safe Drinking Water Act. It is further recommended that this matter be referred to the Office of Special Services with a request that they will conduct an intensive search of the area to determine if there is a local industry that is discharging mercury contaminated wastes.

Respectfully submitted,

William Laffey

Bhupendra Vora

WL&BV:JW:hn Attachments

MONROE TWP. MUA, GLOUCESTER COUNTY TABULATION OF MERCURY RESULTS

SAMPLING POINT

DATE	WEIL #4 RAW DELIVERED	WEIL #5 RAW DELIVERED F	WEIL #6 RAW DELIVERED	SYSTEM - 372 Main St.
The second second	MERC	JRY CONTENT IN ppb		
6-21-76 [W-76]	2.0	1.8	⊌eg.	
7-19-76 [Memo #312]	6.4.	2.8		
8-18-76 [Memo #338]	4.2	1.4		10.8
8-24-76 [Memo #358]	4.2	1.6	∕eg.	1.0
10-5-76 [Routine Insp]	1.4 [total] 1.4* Neg.	Neg. Neg.	leg.	
10-15-76 Well #4 Runs. Well #5 Rested For 12 hours and then took samples	Hr. 0 -1.6 1 -1.6 2 -1.6 4 -2.0	Neg. Neg. Neg.		1.2
At 0, 1, 2, 4, 5 Hours [Mano #738]	5 -1.4	Neg.		
Well 44 Rested For 12 hours and then took samples At 0, 1, 2	0 -6.0 1 -4.0 2 -3.4 4 -3.8	1.0 1.0 0.8		1.6
4, 5 Hours [Memo #438].				

Morganic Mercury.

ESTIVE PROTECTION SERVICE

Remarks_

QUALITY CONTROL LABORATORY

Bacteriological and Chemical Analysis 243 WHITE HORSE PIKE

(609) 428-1303

B7

AUDUBON, N.J. 08106

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Sample take	n from					_	ByJi	m Davi	8				
Condition of	f sample	when dra	wn	5.	7_								
Callegrande 5	Damarka.			_				_					
Date Deliver	red to Lai	boratory	1	<u>2/16</u>	, 19	76	Time					 -	
Condition of	f Sample	upon arr	rival at La	boratory	′ —						<u>_</u>		
<u> </u>	: 			· —————					· -				
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s. Solids							Sulfate						
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erdness							Sulfite	<u> </u>					<u> </u>
ydrogen Sulfide			ļ				Sus. Matter		358	mg/1			
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STROKA SIPPEL MASTELLER

& ASSOC., INC.

ENGINEERING & SURVEYING

ROBERT J. SIPPEL LS. PP EARL H. MASTELLER. PE NELSON L. HOOVER, LS JOHN E. LORENZ

336 ROUTE 70, MARLTON, NEW JERSEY 08053 609.983.7260

December 23, 1976

Mr. Gustav Mihlebach, Superintendent Monroe Municipal Utilities Authority 372 S. Main Street Williamstown, New Jersey 08094

Re: N.J.D.E.P. Special Services
Potable Water Well Survey

Dear Gus:

Pursuant to your request on December 21, 1976, a field crew ran the necessary levels to determine the elevations of wells 4, 5, 6 and the Violet Packing Co. well. The bench marks used were New Jersey Geodetic Control Survey monuments.

Well #4 - Washington Avenue

Well house finish floor elevation 139.25

Top of concrete pump motor base elevation 141.14

Well #5 - Chestnut Street & Water Street

Well house finish floor elevation 165.26

Well #6 - Lake Avenue & Ellen Terrace

Top of concrete pump motor base elevation 144.30

Violet Packing Co. Well - Railroad Avenue

Top of flange of submersible well discharge piping elevation 155.39

Mr. G. Mihlebach, Superintendent Page 2 December 23, 1976

If you should request any additional information or have any questions, please do not hesitate to call me.

Very truly yours,

Robert G. Volk

RGV:nl

MONROE MUNICIPAL UTILITIES AUTHORITY

372 SOUTH MAIN STREET

WILLIAMSTOWN, NEW JERSEY 08094

GUSTAV MIHLEBACH, Superintendent

Phone: 629-4400

December 27, 1976

Amanda M. Miles
Administrative Clerk

Phone 629-4400

30 PA

PR X

2 39

State of New Jersey

Dept. of Environmental Protection

Division Water Resources

P. O. Box 2809

Trenton, N. J. 08625 David Shartant

Attn: Mr. David Longstreet

Supervision, Hazardous Material Program

Re: Mercury Contaminate

Wells # 4, and # 5

3-1415

Dear Dave:

Per your request, enclosed are logs on Wells # 5, #6, and Violet Packing Cannery; also a letter from our engineering firm staking the elevations of the four (4) wells.

An analysis for Mercury performed by Quality ^Control Lab on the sewage treatment plant effluent is also enclosed.

488 360°

Very truly yours,

Monroe Municipal Utilities Authority

Duster mihlebach

Gustav Mihlebach

Superintendent

GM:am

encl: 4

cc: Mr. Volk

file

Fecal Streptococci:MPN/100 ml.

NEW JERSEY STATE DEPARTMENT OF HEALTH STREAM OR WASTEWATER ANALYSIS

Time & Date By Labs	Received	
Lab No		

11

510	Erdina-foundation
PLEASE TYPE OR PRINT WITH BALLPOINT PEN	Date of Collection 2 - 8 - 1927
om Allendaria (n. 1864). 18 maio - Allendaria (n. 1884).	Hour 17 60 A.M. P.M.
Sample No. 18532	Composite Period Cara Interval
Municipality Manager T. S.P.	Collected by
Municipality Manager T. OP. Plant Mores T. Prok. C.	Developed
Stream	
Location PAIL ROAD AVE	Temperature
Description and Remarks:	644
ITEMS CIRCLE	ED BELOW ARE UNSATISFACTORY
Dilutions Requested 10	$\begin{array}{c ccccccccccccccccccccccccccccccccccc$
(Bacteriological)	
	ORATORY RESULTS BACTERIOLOGICAL
Coliform MPN/100 ml ((Confirmed Test); Fecal Coliform MPN/100 ml.

CHEMICAL AND PHYSICAL ANALYSES (mgs./liter, unless otherwise noted)

Color (units)	Chloride	Sulfate	Other Determinations				
Odor (cold)	Suspended Solids	Grease & Oil	MERCURY NEG-				
Turbidity (units)	Ash	Cyanide					
рН	Total Solids	Chromium Total					
Acidity to pH 4	Ash	Chromium Hex.					
Alkalinity to pH 4	Total PO4	Ortho - PO4					
Nitrite N	MBAS	Copper					
Nitrate N	Phenols	Lead					
Ammonia N	COD	Arsenic					
Total Kjel. N	Iron	Zinc					

BIOCHEMICAL OXYGEN DEMAND (mgs./liter)

Field D.O.		Lab.	D.O.		Seed	Require	ed:	Yes		No		
Sample Conc. %	PLEASE CIRCLE	0.1	0.2	0.5	1.0	2.0	5.0	10	25	50	75	100
BOD ₅				••								

	O. W JERSEY STATE DEPA		Time & Date Received 8/11/7
Chem-25 Sept. 75	STREAM OR WASTEW	ATER ANALYSIS	By Labs
	FIELD INFO	RMATION	Lab. No. 7 h 3 8 C
PLEASE TYPE OR PRINT WITH BALLPOINT PEN		Date of Collection	9 -// - 1977
•		Hour 10 + 15	A.M P.M
Sample No. 20636		Composite Period G	AB_ Interval
Municipality MONROF Plant VIOLET PAC	TWP	Collected by	WELT PATTERSON
Plant VIOZET PAC	CKINC	Developed	
Stream		Flow Rate	DISTRICT
Location		Temperature	
Description and Remarks:	Pond	(ATRAY, ON	LACONSP1 6 1977
			State of New Jersey Dent. Environmental Protection Division Water 1
	ITEMS CIRCLED BELOW A		ater Resources
Dilutions Requested (Bacteriological)	10 1 1	0.1 10.2 10.3 10.4	. 10.5 10.6
	LABORATORY BACTERIOL		
Coliform MPN/100 ml.	24000+ (Confirmed	Testy, Fecal Coliform MPN	1/100 ml. 3500.
Fecal Streptococci:MPN/100 ml.	2400 t	Other	

CHEMICAL AND PHYSICAL ANALYSES (mgs./liter, unless otherwise noted)

	·		
Color (units) ND	Chloride 64	Sulfate 42	Other Determinations
Odor (cold) VD	Suspended Solids 66	Grease & Oil 104.0	SODIUM 310,
Turbidity (units) 50	(Ash 24	Cyanide	POSASSIUM 47.
V pH 9.5	2 Total Solids 1180	Chromium Total 045	CADMIUM
Acidity to pH 4	Ash 904	Chromium Hex. NJ	0.002
Alkalinity to pH 4	Total PO ₄ 6.9	Ortho - PO4	
Nitrite N 0.055	V MBAS < 0.3	Copper	
Nitrate N N D	Phenols	Lead	DIMCION OF LIBORA
Ammonia N 4.0	COD 360	Arsenic	DIVISION OF LABORATORIES ANALYSIS COMPLETED
Total Kjel. N 17.9	Iron 2.8	Zinc 0,48	1774 710//

BIOCHEMICAL OXYGEN DEMAND (mgs./liter)

REPORT SUBMITTED

Field D.O.		Lab.	D.O.	0	Seed	Requir	ed:	Yes		No		
Sample Conc. %	PLEASE CIRCLE	0.1 (0.2	0.5	1.0	2.0	5.0	10	25	50	75	100
≫80D ₅			<		Y —		137					

	STATE DEPARTMENT OF HEALTH OR WASTEWATER ANALYSIS Time & Date Received \$\frac{\xi_1\cdot 7}{3\xi_5\sqrt{5}}\$ Lab. No. Th 3\xi_5\sqrt{5}
FIE	ELD INFORMATION
PLEASE TYPE OR PRINT	Date of Collection $8 - 11$ 1977
WITH BALLPOINT PEN	
- 21 21	Hour //: 10 A.M. P.M.
Sample No. 20634	Composite Period Interval
Municipality Monkow TWP Plant VIOLET PARCING	Collected by MENNEL & PATTECS Residual Chlorine: Immediate
Plant VIOLET PARCHO	Developed
Stream	Flow Rate
Location	Temperature
Description and Remarks:	PH ADJUSTED.
AND SCREENED	(EFFLUENT TO GC SA)
ITEMS CIRCLE	D BELOW ARE UNSATISFACTORY
Dilutions Requested 10	1 10.1 10.2 10.3 10.4 10.5 10.6
(Bacteriological)	
LAB	ORATORY RESULTS

BACTERIOLOGICAL

State of New Jersey

Dept. Environmental Protection

CHEMICAL AND PHYSICAL ANALYSES (mgs./liter, unless otherwise notice to a protection with the protection of the protection

SEP 1 6 1977

24,000 + (Confirmed Test) Fecal Coliform 11 100 ml

	: 		<u>/</u>			. 1911	M Water Resources
	Color (units) ND		Ehloride 15	V	Sulfate 85		Other Determinations
	Odor (cold) IV D		Suspended Solids 520	1	Grease & Oil 67,4		Sobjum 22.
	Turbidity (units)200	/	Ash 212		Cyanide	_	POTASSIUM 60
	pH 4.5		Total Solids 1216		Chromium Total 005	/	CADMIUM 0.003
	Acidity to pH 4	7	Ash 488	\checkmark	Chromium Hex. ND		
	Alkalinity to pH 4	V	Total PO4 /0,0		Ortho - PO4		
	Nitrite N 0.025		MBAS < 0.3		Copper		DIVISION OF LABORATORIES
V	Nitrate N 3.5		Phenols		Lead		AMALYSIS COMPLETED
	Ammonia N 11,1		COD 1240		Arsenic		SEP 1 2 19/7.
V	Total Kjel. N 39.8		Iron 8.0		Zinc 0.75		DEPONT CHRMITTED
							אבורטונו סטטוווורטי

Fecal Streptococci:MPN/100 ml._

BIOCHEMICAL OXYGEN DEMAND (mgs./liter)

Field D.O.		Lab.	D.O.	0	Seed	Require	ed:	Yes	\supset	No		
Sample Conc. %	PLEASE CIRCLE	0.1	(0.2)	0.5	1.0	2.0	5.0	10	25	50	75	100
BOD ₅			4 -		651	>-						

Executive Offices: Water Research Bullding 44 Sintsink Drive East Port Washington, New York 11050 Phone 516 8N3-6760 Cable WATER

Geraghty & Miller, Inc.

CONSULTING GROUND-WATER GEOLOGISTS AND HYDROLOGISTS

Mr. Gustav Mihlebach Monroe Utilities Authority 372 S. Main Street Williamstown, New Jersey

Dear Sir:

Enclosed are the water quality analyses of the samples collected from your wells in January. This work was done as a part of the Delaware Valley Regional Planning Commission study of ground- and surface-water contamination being conducted in Gloucester, Camden, Burlington and Mercer Counties.

Thank you for your cooperation.

Sincerely,

GERAGHTY & MILLER, INC.

Paul Roux

PR:am Enclosure



PENN ENVIRONMENTAL CONSTITUTIONS INTO ANTS, INC. FORT PITT PROFESSIONAL BUILDING 1517 WOODRUFF STREET PITTSBURGH, PA. 15220 412-381-1133

WATER ANALYSIS REPORT

CLIENT		WATER ANALT	313 NEFORT	PEC PROJECT NO.	\ 			
COAN	٠	Marcht 4	Milly	253-	333P			
SAMPLE SOURCE TIME AND DATE		Geragety w	11944					
DYRPC MOIS	CHEMIST	Will 4	SP SAMPLE	SAMPLE NO.				
2-17-77	<u> </u>	8n	·	25 335				
BASIC PARAMETER	S	NUTRIE	NTS	ORGANICS				
pH-Units	49	NO ₂ -N mg/l		TOC mg/l				
ALK-TOT mg/l as CaCO ₃	1/2	NO ₃ -N mg/l	3./	COD mg/l				
ACID mg/l as CaCO ₃		NH ₃ -N mg/l		BOD _s mg/l				
pH _a HOT mg/i as CaCO ₃		PO ₄ -P ortho mg/l		CN-TOT mg/l				
COLOR Pt-Co Units		POP total mg/l	.03	CN-AMENABLE mg/l				
TURBIDITY JTU		TKN mg/l		OILS-EXT mg/l				
SO ₄ mg/l	</td <td>META</td> <td>LS</td> <td>PHENOL mg/i</td> <td></td>	META	LS	PHENOL mg/i				
SP COND. µmhos/cm		Al mg/l		MBAS mg/I				
HARD-T mg/l		Cd mg/l	Ka1	BACTERIOL	OGICAL			
Ca mg/l	24	Cr mg/l	K01	TOT-COLIFORM colonies/100 ml				
Mg mg/l	70	Cu mg/l	11	FECAL-COLIFORM colonies/100 ml				
CI mg/I	/3	Fe-TOT mg/l	16	FECAL STREP colonies/100 ml				
F mg/l		Fe-DISS mg/l		ADDITIO	NAL			
SOLIDS	·	K mg/l	20	Co mall	K.al			
TOTAL mg/l @ 103 ⁰		Mn mg/l		Hot mall	1 004			
TOT VOL mg/l @ 550°		Na mg/l	7.1	as mall	K.03			
SUSP mg/l @ 1030		Ni mg/l	K.03					
SUSP-VO L mg/l @ 550°		Pb mg/l	K.03	(12 m)				
DISS mg/l @ 180°	95	Si mg/l						
SETTLE ml/l		Zn mg/1	30	·				
REMARKS:				·				



PENN ENVIRONMENTAL CONTRACTOR ANTS, INC. FORT PITT PROFESSIONAL BUILDING 1517 WOODRUFF STREET PITTSBURGH, PA. 15220

412-381-1133

WATER ANALYSIS REPORT

CHENT		==	===	=				==	<u> </u>		PEC PROJECT NO.	_	=	==		
					PROSTE & M	100	7,5				35.3	_ ;	7	?-24	P	
SAMPLE SOURCE TIME AND DATE		مذ			The state of the s	w	<u> /</u>						تد			
DATE RECEIVED		HE	7/ 4181	20	D Will 5	FSAM	<u> </u>				SAMPLE NO.	_				
2-17-77			8	N	Geragety & M.						353	3	<u>6.</u>			
BASIC PARAMETE	RS				NUTRIENTS				ORGANI	CS						
pH-Units	,,		4	8	NO ₂ -N mg/l						TOC mg/l	\int		I		
ALK-TOT mg/l as CaCO ₃		\perp	1	2	NO ₃ -N mg/l				2	Z	COD mg/l					
ACID mg/l as CaCO ₃		\perp			NH ₃ -N mg/l						BOD ₅ mg/l					
pH _a HOT mg/l as CaCO ₃					PO ₄ -P ortho mg/l						CN-TOT mg/l					
COLOR Pt-Co Units					PO ₄ -P total mg/l				0	2	CN-AMENABLE mg/I					
TURBIDITY JTU					TKN mg/l						OILS-EXT mg/l					
SO _s mg/l			1	2	METAL	.s					PHENOL mg/I	1		\perp		
SP COND. µmhos/cm					Al mg/i						MBAS mg/l					
HARD-T mg/l					Cd mg/l			ζ.	0	1	BACTERIOLO	GIC	AL			
Ca mg/i			3	3	Cr mg/l			٧,	0	/	TOT-COLIFORM colonies/100 ml	T	T	Τ		T
Mg mg/l			3	4	Cu mg/l			4	0	/	FECAL-COLIFORM colonies/100 ml	7		T		T
CI mg/I			/	6	Fe-TOT mg/l				Ż	3	FECAL STREP colonies/100 ml			T		
Fmg/l					Fe-DISS mg/I						ADDITION	AL				
SOLIDS					K mg/l				2.	4	Co mall		\int	\prod		d
TOTAL mg/l @ 103°					Mn mg/l						Mg/ mg/l				0	0
TOT VOL mg/I @ 550°					Na mg/l				9.	8	as mall					a
SUSP mg/l @ 103 ⁰					Ni mg/l			۷,	0	3	A -					
SUSP-VO L mg/l @ 550°					Pb mg/l					3						
DISS mg/l @ 180°			2	6	Si mg/l							\int				
SETTLE mi/I					Zn mg/l			<	0	/			\int			
REMARKS:									•				-	,		
											·					



- PENN ENVIRONMENTAL COLLINATION. FORT PITT PROFESSIONAL BUILDING 1517 WOODRUFF STREET PITTSBURGH, PA. 15220 412–381-1133

WATER AÑALYSIS REPORT

CLIENT			<u> </u>	4.0				PEC PROJECT NO.	====	=			
SAMPLE SOURCE TIME AND DATE			Buch	7 × 1/2	<u>elu</u>				<u>3-</u>	3	3	28	
DATE RECEIVED		M	Skrops nrse)	t [1]000 lo	á	2/1	/	•					
DATE RECEIVED	CHEM	IST /		TYPE OF SA	MPLE	~		SAMPLE NO.					
2-17-77		151	<i>r</i>	<u></u>	· 			25.3	<u>3Z</u>	_	===		
BASIC PARAMETE	RS			NUTRIENTS				ORGA	VICS	•			
pH-Units		44	NO ₂ -N mg/l	•.				TOC mg/l					
ALK-TOT mg/l as CaCO ₃		10	NO ₃ -N mg/l				98	COD mg/l					
ACID mg/l as CaCO ₃			NH ₃ -N mg/l					BOD _s mg/i					
pH ₈ HOT mg/l as CaCO ₃			PO ₄ -P ortho mg/l					CN-TOT mg/l					
COLOR Pt-Co Units			PO ₄ -P total mg/l				03	CN-AMENABLE mg/l					
TURBIDITY JTU			TKN mg/l					OILS-EXT mg/I					
SO, mg/i		25		METALS				PHENOL mg/l					
SP COND. µmhos/cm			Al mg/l					MBAS mg/l					
HARD-T mg/l			Cd mg/l			<	01	BACTERIO	.0G	ICA	\L		
Ca mg/l		4.0	Cr mg/l			11	ol	TOT-COLIFORM colonies/100 ml		\prod			Π
Mg mg/l			Cu mg/l				05	FECAL-COLIFORM colonies/100 ml					
CI mg/I		10	Fe-TOT mg/l				40	FECAL STREP colonies/100 ml					
F mg/i			Fe-DISS mg/l					ADDITIO	ONA	,L		•	
SOLIDS			K mg/l				1.6	Co night				₹,	01
TOTAL mg/l @ 1030			Mn mg/l					No mall			<u> </u>	00	05
TOT VOL mg/I @ 550°	-		Na mg/l				3.5		\prod		$\overline{\cdot}$		05
SUSP mg/I @ 1030			Ni mg/l				03	. 0"					
SUSP-VOL mg/l @ 5500			Pb mg/l				03						
DISS mg/l @ 180°		56	Si mg/l										
SETTLE mi/l			Zn mg/l			٦	37						
REMARKS:													
			······································										

B17

lunicipality	and the first of the second states	<u> </u>		LE Memo # 5	GO:	County Date Collect Project # W	ed 2/4/	197
ACTERIOLOGICAL	ANALYSES:	Coliform organis Chlorine residua	sms determined bils are reported in	y the membrane f	ilter technique	are reported in c	olonies per	100 ml.
Sample Number		Point	of Collection			rm Organisms	Chlorine	
12003	i ilkova i a priza a se se con	- (位)をひか (AA) ションの	Market St. Contraction of the State	A CAN	Fecal	Total	Free	Total
	123/201	1010						
er saendi								
	WILL	7	Rawle	ite.	19 19:30			
	Indle 1	A STATE OF STATE OF STATE OF	Wall-Select	i englemente				
32005	WHIH.	4 = 1 m	faw b	We .				t yanga ha
	In a late	Am Use						
						Part Carlo		10 gen
			'ATSNIA WATER ST		TIONAL INTERIM: P	rimary Regulation	IS. 35 Sec. 37 - 3	The state of
emple (fumber) & i	ALCOHOL: MANY			32005	7	Sample Number		03
30 (10).					7			03
100	No. Art				7	Sample Number Arsenic (0.05)		03
ctor,	No. Art				7	Sample Number Arsenic (0.05) Barium (1.0)	320	03
ctor,	No. Art				7	Sample Number Arsenic (0.05) Berium (1.0) Cadmium (0.010)	320	03
do (III) do (III) bidity (5)	No. Art				7	Sample Number Arsenic (0.05) Berium (1.0) Cadmium (0.010) Chromium (0.00)	320	03
dor (III) roidity (5) Rainity to pH 4	No. Art				7	Sample Number Arsenic (0.05) Barium (1.0) Cadmium (0.010) Chromium (0.00 Copper (1.0) Cyanide (0.20) Lead (0.05)	320	03
doc (III) doc (III) abidity (5) Relinity to pH 4 trate as NO ₃ (45)	No. Art				7	Sample Number Arsenic (0.05) Berium (1.0) Cadmium (0.010) Chromium (0.00 Copper (1.0) Cyanide (0.20) Lead (0.05) Mercury (0.002)	320	MD
dor (IIII) spidity (5) kalinity: 19, pH 4 itrata as NO ₃ (45) that Dis Solids (500) sS/CAS (0.5)	No. Art				7	Sample Number Arsenic (0.05) Barium (1.0) Cadmium (0.010) Chromium (0.00) Copper (1.0) Cyanide (0.20) Lead (0.05) Mercury (0.002) Setenium (0.01)	320	ND
dor (IIII) abidity (5) Ralinity to pH 4 itrata as NO ₃ (45) uoride 25 (250) stal Dis Solids (500) S/LAS (0.5)	No. Art				7	Sample Number Arsenic (0.05) Barium (1.0) Cadmium (0.010) Chromium (0.01) Cyanide (0.20) Lead (0.05) Mercury (0.002) Setenium (0.01)	320	MD
color (10), color (III) color	No. Art				7	Sample Number Arsenic (0.05) Berium (1.0) Cadmium (0.010) Chromium (0.01) Cyanide (0.20) Lead (0.05) Mercury (0.002) Selenium(0.01) Silver (0.05)	320	M.D
dos (IIII) abidity (5) Realimity to pH 4 itrata as NO3 (45) tal Dic Solids (500) stal Hardness (150) tal Iron (0.3) regness (0.05)	No. Art				7	Sample Number Arsenic (0.05) Berium (1.0) Cadmium (0.010) Chromium (0.00 Copper (1.0) Cyanide (0.20) Lead (0.05) Mercury (0.002) Selenium(0.01) Silver (0.05) Phenol (0.001)	320	M.D
dor (IIII) rbidity (5) Ralinity to pH 4 itrate as NO ₃ : (45) tall Dis Solids (500) SS/CAS (0.5) tall Hardness (150) tall Hardness (150) dat fron (150)	No. Art			32005 3 5.088 35.000	7	Sample Number Arsenic (0.05) Barium (1.0) Cadmium (0.010) Chromium (0.00 Copper (1.0) Cyanide (0.20) Lead (0.05) Mercury (0.002) Selenium(0.01) Silver (0.05) Phenol (0.001) Endrin (0.0002)	326	MD
der (IIII) sbidity (5) Ralinity, 10, pH 4 Itrata, as NO ₃ (45) uoride 2; (250) tal Dir Solids (500) ss/LAS (0.5) dal Hardness (150) cal Iron (0.3) engensse (0.05)	No. Art				7	Sample Number Arsenic (0.05) Barium (1.0) Cadmium (0.010) Chromium (0.01) Cyanide (0.20) Lead (0.05) Mercury (0.002) Selenium (0.01) Silver (0.05) Phenol (0.001) Endrin (0.0002) Lindane (0.004)	320	MD
der (IIII) abidity (5) Ralinity to pH 4 trate as NO ₃ (45) that Dis Solids (500) RS/CAS (0.5) tal Hardness (150) das Iron (0.3) ergensses (0.05) cas Iron (0.3) phase (50)	No. Art			32005 2 5.09eB 25.009 2	7	Sample Number Arsenic (0.05) Berium (1.0) Cadmium (0.010) Chromium (0.010) Copper (1.0) Cyanide (0.20) Lead (0.05) Mercury (0.002) Selenium(0.01) Silver (0.05) Phenol (0.001) Endrin (0.0002) Lindane (0.004) Methoxychlor (0.1	320	MD
dos (IIII) sbidity (5) Realinity to pH 4 itrate as NO3 (45) tal Dic Solids (500) stal Dic Solids (500) stal Hardness (150) dai Iron (0.5) dair (0.5)	No. Art			32005 2 5.09eB 25.0000 /	7	Sample Number Arsenic (0.05) Barium (1.0) Cadmium (0.010) Chromium (0.01) Cyanide (0.20) Lead (0.05) Mercury (0.002) Selenium (0.01) Silver (0.05) Phenol (0.001) Endrin (0.0002) Lindane (0.004)	320	MD

MEMORANDUM

STATE OF NEW JERSEY DEPARTMENT OF ENVIRONMENTAL PROTECTION

TO:	Robert	L. Vincent	thru Joseph	Μ.	Mikulka

FROM: William 57 Mennel, Jr. and Paul F. Tomkavage

SUBJECT: Violet Packing Company, Monroe Township

Gloucester County

DATE: March 15, 1977

On February 8, 1977 we conducted an investigation of the Violet Packing Company. During this investigation we met with Mr. James V. Sclafani, the President of the Company, who answered our questions and showed us around the Plant. He explained that their principal raw material is tomatoes. The tomatoes as they come in are washed and then cooked and processed into tomato sauce. The wash water and the process wastewater are treated in their pre-treatment plant before being discharged into the Monroe Township Sewerage System. The firm operates and discharges only during the months of July, August and September. The. discharge is restricted to the hours of 12:00 a.m. to 4:00 a.m. The pre-treatment system consists of a bar screen, a tank for pH adjustment, a tank for the addition of alum, three primary settling lagoons, an aeration lagoon and a final settling lagoon. The primary settling lagoons have rubber liners but the liners are torn and have holes in numerous places. The aeration lagoon and final settling lagoon are unlined. Mr. Sclafani informed us that he is having problems with vandalism at the plant.

After our inspection of the Violet Packing Company we met with Mr. Gustav Mihlebach, Superintendent for the Monroe Municipal Utilities Authority. He informed us that although Violet Packing is tied into the Municipal System, they did not discharge into the system during the 1976 season, but had been discharging to a field behind their plant. This information was based on his knowledge of the Monroe M.U.A. treatment plant's flows. Mr. Sclafani, however, informed us that the only time he discharged to the field behind his plant, was when the pump motors burned out.

At the request of Richard Dalton, of Special Services, we took a sample of Violet Packing's well water. The requested parameter for this sample was for Mercury. The result of this analysis was negative.

A39:MC

MEMORANDUM

STATE OF NEW JERSEY DEPARTMENT OF ENVIRONMENTAL PROTECTIO

TO: Richard Dalton	
FROM: Robert L. Vincent	MINE 4 TO 4077
SUBJECT: Violet Packing Company, Monroe Township Gloucester County	MAK 1 7 1977

On February 8, 1977 Basin Personnel inspected the above referenced facility. This inspection revealed the existence of two unlined lagoons and three lined lagoons, which are in a state of disrepair, for the pretreatment of food processing wastewater prior to discharge into the municipal system. The plant is "down" now and use of these facilities is not expected to be resumed until processing starts again sometime early this summer.

I would appreciate comment from you concerning the use of the unlined lagoons for this type of wastewater.

A39:MC

M. M. U. A. Water Supply System

L'en from NJDEP M Wilford IR: Naw Water Tests on # 4 and # 5 Welled Mercury levels

11-12-76 Meeting with me Clinori Re: New Well, Mercury Content

Meeting with Auditor Mr. Trightinger E: Financial Condition of M.M.U.A. General Fund

11-16-76 Meeling with John Neider Erme Condeson, Havry Claentyer AC Schultes C: Cost Figures New 1000 GPM Wall Discussed Mercury content

11-18-76 Grat samples - Caw Water from Wells # 4 and # 5 debreist to
Technological Services, Cameden
Luckt Lat Owendon
C: Tools on Mercury

Chromium Copper

11-24-76 The to NIDEP Mr John Wieford

Ce: Progress and steps taken to date regarding

mer own levels mer cury levelo

Joe Miller NJDEP Water Resources (Pollution)
Bei Steps taken on Mercury
Bei 11-24-76 Callet

11-29-76 Meeting with our Engineering Firm Sot Volk Earl Mesteller Re: Mercung in Wells and Budget Figures 12-1-76 Meeting with Mr. Denovi R: Progress report on Wells Quality Control Laboratory, audubon liken on 11-18-76 #5 less den 0.5 0.5 Lead 0.10 X 0.26 0.12 Zine less then 0.03 Chromeum 0.03 0.001 12-3-76 Technologist Resources, Inc. Camber Test Césults laken on 11-18-76 # 5 0:01 Copper 0.14 0. 005 0.005 dead 0.05 0.005 Zuc 0.02 < 0.02 Chromum 0.0021 Mercury 12-10-76 Meeting with Gus Schulles Jr., Don Jangus, Bot Volk at Schulles of mercung in Welleamstown area.

Clot Wells of various depths in area.

M. a. Kocate the massive concentration

Can new well be located it base of

new elvated Tower

Cohansey Strata - Land

Test wells

Baz

609) 468-3396

VAL ASSOCIATES

Plating Analysis & Consulting For Electronics Industry Water, Air & Soil Analysis

Glovest MUA Monro MUA P.O. Box 162

PHILIP V. DATZ, JR. Chemist

748 Ridge Drive Road Mantua, New Jersey 08051 June 9, 1978

Water Analysis Samples

5/26/78 Sample Taken Mercury Filtered - .4 pps 5/26/78 Sample Taken Mercury Unfiltered 4 ppb

Well #4

5/8/78 Sample Taken

Mercury -.35 ppb 5/10/78Sample Taken (Well #

- 1.6 ppb Mercury

June June	PWS TO NO. OBNOCZ
Water 3	upply Purveyor Monroe Municipal Utilities Authority Date March 20, 1980
Municip	Monroe Township County Gloucester
Mailing	Address 372 South Main Street, Williamstown, New Jersey 08094
Admini	Strator <u>Mr. J.V. Dinovi, Chairman</u> Lic. Operator: T Gustav Mihlebach W same
Busines	s Phones: 609-629-4400 609-589-2976 (home)
Person 1	nterviewed Mr. Gustav Mihlebach Position Superintendent
1.	Source: Location, Description, Capacity (mgd) Washington Ave. Well # 4-0.72 mgd (for emergency use
	only), Chestnut Street & Water Street Well # 5-0.81 mgd, Ellen Terrace & Lake Ave.
,	Well # 6-0.60 mgd, corkery Lane Well # 7-1.14 mgd.
	Est. Total Effective Cap. (mgd) 3.27
2.	Treatment: Wells # 4,5, & 6-Chlorination (gas) and pH adjustment with Caustic Soda.
	Well # 7-chlorination (gas) and pH adjustment with lime.
	Est. Total Effective Cap. (mgd)
3 Fi	Chestnut & Water Storage: Description Capacity (MG) Chestnut & Water St. elevated tank-0.15 mg, Herbert
0. 1	Blvd. elevated tank-0.30 mg, Corkery Lane & Black Horse Pike elevated tank-1.0 mg Est Total Effective (MG) 1.45
4.	Auxiliary Power Well # 4-gasoline engine direct drive, Well # 5-diesel engine, direct drive, Well # 7-diesel generator for well & treatment. Total 2.67 mgd
5.	Emergency Interconnections none Max Day 1.63(10/25/79) Min. Day .124 (1/31/80)
6.	Plant Delivered (mgd): Maximum 1.022 (7/79) Minimum .520 (3/80) Annual Average .801
	Bulk Purchase Fromnone
	Bulk Sale To mg
7	Number of Services 2,699 % Metered 100 Total Est. Population Serviced 9,400
8.	Municipalities served and est. services in each Monroe Twp
9.	Distribution Mains: Size 2 to 12 ins. Pressures 40 to 50 psi. Fire Hydrants yes
10.	Water restrictions
11.	New Construction & Project #
12.	Plant Chemical Physical Analysis (type, freq.) Chlorine residuals twice a day; pH is checked daily
13.	Monthly becteriological sampling of system by purposes. No serviced 10 No also 10
- -	Monthly bacteriological sampling of system by purveyor: No. required 10 No. taken 1
	Name of Laboratory

NEW JERSE PARTMENT OF ENVIRONMENTAL JECTION ROUTING INSPECTION REPORT - PUBLIC WATERSUPPLY

ater Supply Purveyor Monroe Municipal Utilities Authority	Date March 20, 1980
Deficiencies noted: Source Well # 4-well blow-off is not_protected against entry of p	ollution. Undergro
gasoline storage tank is located within 100 ft. of well. 2. Well #	4-past Brueau record
indicates raw water from this well contains mercury in excess of the	MCL established by
Safe Drinking Water Act. Treatment -	
	•
Storage and Distribution 1. Distribution system contains undersized mains.	
Other	
Sampling: Dates/Comments 3/20/80 microbiological analysis was satisfactory analysis showed the finished water from Well # 7 to be corrosive. Land	
Condition of facilities Excellent [x] Good [] Fair [] Poor [] Uns	tatisfactory [] tatisfactory [] tatisfactory []
Specific Comments	
Operator's Reports Satisfactory	
Immediate Requirements* 1. Use Well # 4 in an emergency only. 2. Use We	ell # 5 only when
necessary to meet daily water demands. 3. Adjust pH of finished water to render water less corrosive	er from Well # 7
	•
	- 1 · · · · · · · · · · · · · · · · · ·
General Recommendations 1. Maintain surveillance to detect leakage of gas 2. Gradually replace undersized mains smaller than 6 in. pipe diame	
OTE: Kindly inform this Department of your actions relative to implementation of item 17 Inspected: Signature Signature William J. L. Name	5/9/80
enior Environmental Engineer Supervising F _{nvironmental Eng}	incer

· Monroe Municipal Utilities Authority

372 South Main Street Williamstown, New Jersey 08094

Gustau Mihlebach Superintendent Uhone: 629-4400 RECEIVED acqueline Schoenewald

Office Manager

AUG 7-80

Dhone: 629-4400

August 5, 1989 STATE DEPT. OF ENVIRONMENTAL PROTECTION BUREAU OF POTABLE WATER

State of New Jersey Department of Environmental Protection Division of Water Resources P.O. Box CN-029 Trenton, New Jersey 08625

Attention: Mr. Daniel S. Mozer

Senior Environmental Engineer

Inspection Report

March 20, 1980

Dear Mr. Mozer:

In the Inspection report - under number 14 deficiencies noted:

- (1) Well #4 blow off has been screened to prevent entry of any foreign matter.
- Well #4 is not being used, the monthly report to N.J.-D.E.P. is marked off line.

Under number 17 Immediate requirements:

- Well #4 is Off Line. (1)
- .(2) Well #5 is being used to meet the heavy daily
 - (3) We are increasing the percentage of lime in the slurry solution to adjust the pH.

Very truly yours,

MONROE MUNICIPAL UTILITIES AUTHORITY

GUSTAV MÍHLÉBACH

Tuster mihebach

Superintendent

GM/cdf cc: File

	TABU	LATION OF AN	IALYTICAL DA	OF ENVIRONI	IC TER SUP	PLY	
(inlepality	£. • .	E TWP.	TILITIES	AUTHORITY		County EL	3-13-8
ollected by	Buan	Satisfied Annual Control			-67	Project # W-	\$ 3
ACTERIOLOGICAL	ANALYSES:		ms determined b	y the membrane fil	SATE OF STREET	reported in co	onies per 100 ml
Sample Number		Point	of Collection	Asia Walana			Chlorine Residua
STATE OF STA	Vic. 4	と 養食者 (がなる) 123		RAW WATER	9:45AM	Total	Free Total
₹0638\$* ¹	all and the second of	CORKE	ALCOHOLD AND AND AND AND AND AND AND AND AND AN	The second of th	10:20		
		WELL AV			71:00	//	
06387	the second of th	and the second of the second o	all the second of the second o	The state of the s	1/140 AM	1/3	1
2 - 10 - 2 - 2 - 2 - 2 - 2 - 2 - 2 - 2 - 2 -	A CONTRACTOR AND ADDRESS OF THE PARTY AND ADDR					Jan Alexander	
Sycalor				. N. 7		人生美	
				Tanahara (
HYSICAL — CHEMI	CAL ANALYS			except color, odor andards and/or Nati			
		Tarana a					T - 4
Sample Number			06384	Sample Number	06385	06386	06387
Color (10)				Arsenic (0,05)			
Ødor (III)				Barium (1.0)	The state of the s		
Turbidity (5)				Cadmium (0.010)			
pH	The state of the s		,169	Chromium +6 (0.05)			The same of the sa
Alkalinity to pH 4				Copper (1.0)		<u> </u>	
Niffate as NO3 (45)				Cyanide (0.20)			
Chloride (250)			10 00D 7	Lead (0.05)	10.0005 K	10.0012	0.0044
Tetal Dis, Solids (500)			10.0007	Mercury (0.002) Selenium (0.01)	0.0003	70,0012	0.0077
AB\$/LAS (0.5)				Silver (0.05)			
Total Hardness (150)				Special Control			
Total Iron (0.3)				Phenot (0.001) Endrin (0.0002)			
Manganese (0.05)				Lindane (0.004)			
Su phate (250)	167 - 4-5			Methoxychlor (0.1)			2.024
Fjuoride (2.0)				Toxaphene (0.005)			
Line (5.0)				2.4-D (0.1)			SENTE
			10.000	Silvex (0.01)	ALC: NO.	7 75	JEINE
				74. N. C. C. C.	96.4		+ ve 1988
						E STATE M	AY SE
				434.3.123			3.00
	· # _ *				1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1	3.47 ³ .c	
	the state of the s	The state of the s		AT TIME		萨里克里尔 斯里	areas Trees and The
STATE WELL	也带件。	UMPED	to Wast	= 2 HRS of	RIOR TO	SAMPLIN	
		and the second	HEET 2 - WA	TER PURVEYOR			
TO WELL COOK	الماليكي من	ADTAL .					FOR BUILDING

(BOTABLE WASTER)

, Monroe Municipal Utilities Authority

372 South Main Street Williamstown, New Jersey 08094 MAR 1 7 198

RECEIVED

Gustau Mihlebach Superintendent

Uhone: 629-4400

March 15, 1982

NJ. STATE DEPT. OF ENVIRONMENTED PROTECTION
Jacqueline Schoenewald
Executive Director

Phone: 629-4400

N.J. State Dept. Environmental Protection P.O. Box 2809

Trenton, New Jersey

Attention: Mr. Dan Mozer

RE: Groundwater Quality

Dear Dan:

To confirm our telephone conversation on Friday (3/12/82, 3:30 P.M.), I would like to mention the subjects discussed.

#4 Well, whether it was abandoned, capped, sealed, amount of Mercury in the water, gasoline seepage from the old Petes' Getty Gas Station and is it being used. The Well is off line, however, I pump it overboard to keep it fresh, just in case of an emergency, such as a fire. A new chlorine booster pump will be installed in the next few weeks.

#5 Well, that it is supposed to be a marginal well and only used in an emergency because the Mercury Tests years back showed 1.5 ppb. still under the E.P.A. standard of 2 ppb. The Well is now pulled for general maintenance and the repair work will be bidded within 2 weeks.

Groundwater Quality in New Jersey. An inspection of Toxic Contaminants, March 1981 by Robert K. Tucker, PhD.

I asked whether the laboratory tests performed in the aforementioned study reflected any serious problems with our wells and the statement given, that I could make to the M.M.U.A. at its meeting on Tuesday, March 16, 1982 is "The quality of our potable water shows no significant amount of pesticides, nothing to be alarmed over."

Mercury tests had been performed by Quality Control Laboratory 2/1/82 at #5 Well, Test Well on Avery Drive and #7 Well and the results were less than 0.002/mg/l at all three locations. Also the State may be down in the near future to perform an inspection.

Very truly yours

Justin Mihlebach

GM/cdf cc: File

PW-2/NW	
May 77	

NEW JERSEY S

: DEPARTMENT OF ENVIRONM

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∟ PROTECTION

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TABULATION OF ANALYTICAL DATA FROM PUBLIC WATER SUP	PLY
SUPPLY MONROE MUNICIPAL LITE AUTHORITY	County GLOUCESTER
Municipality MONROE TWP	Date Collected 3-23-82
Collected by Suain Keumph 201582 Memo # 82-67	Project # W-
BACTERIOLOGICAL ANALYSES: Coliform organisms determined by the membrane filter technique are	reported in colonies per 100 ml

Chlorine residuals are reported in ppm. "

		Goliform C) rganisms	Chlorine Residual		
Sample Number	Point of Collection	Fecal-	Total	Free	Total	
06384	WELL # 6 LAKE AVE (RAW WATER)	9:45AM		0	0	
06385	WELL #7. CORKERY LN.	10:20				
06386	TEST WELL AVERY DR.	11:00				
06387	WELL #4 WASHINGTON ST. 4	11:40 AM		\	V	
			/			

PHYSICAL - CHEMICAL ANALYSES: Determinations are in ppm except color, odor, turbidity, and pH. Figures in parentheses are from the N. J. Potable Water Standards and/or National Interim Primary Regulations.

		 			ary riogarations.	·
Sample Number		06394	Sample Number	06385	06386	06387
Cojór (10)			Arsenic (0.05)			
Odor (III)	·		Barîum (1.0)			
Turbidity (5)			Cadmium (0.010)			
рН			Chromium ⁺⁶ (0.05)			
Alkalinity to pH 4			Copper (1.0)			
Nitrate as NO ₃ (45)			Cyanide (0.20)			
Coloride (250)			Lead (0.05)			
Total Dis. Solids (500)	· .	10.0007	Mercury (0.002)	10.0005 K	10.0012	70.0044
ABSTLAS (0.5)	•		Selenium(0.01)			
Total Hardness (150)		,	Silver (0.05)			
Total Iron (0.3)			Phenoi (0.001)		•	
Manganese (0.05)	•		Endrin (0.0002)			·
Sodium (50)			Lindane (0.004)			
Sulphate (250)		·	Methoxychior (0.1)		REPORT S	USMITTED
Fluoride (2.0)			Toxaphene (0.005)		ADD 9	0 1982
Zinc (5.0)			2.4-D (0.1)		,	_
			Silvex (0.01)		NUDOH En	vironmental Laboratory
					Chemistry	ECHOICE Y

WELL# 5 OUT OF SERVICE AT TIME OF SAMPLING
WELL# 4 PUMPED TO WASTE 2 HRS PRIOR TO SAMPLING

* Exceeds standard SHEET 1 - BUREAU OF POTABLE WATER

Reg file	- Marrai Mit, Glow	Com That the second
Form VST- 001 '.	STATE OF NEW JERSEY	CHAIN OF CUSTOD
PLEASE TYPE OR PRINT	Department of Environmental Protection Water Analysis	BACT. LAB NO.
MUNICIPALITY	STREPS A	DATE REC'D.
FACILITY LOCAT	Mulle 322 a Washing It	BOTTLE NO. 1464
REPRESENTATIVE TITLE	COLL NAME	DATE REC'D
REMARKS Egros Januliis	under enful storm	STORET READ
Culvet from I'll	Will By selflan	READ
Station Identification Numb	BYR, MO. DAY HOUR	Sample No.
=	ाना । स्विकालका वर्ष विकास	P8.
	THE EPPPP UPPP	<u> </u>
FIELD ANALYSIS	BACTERIOLOGICAL - DILUTIONS (REQUESTED) Fecal Coliform	□p ^H (LAB) (39) P00403,
Water (2) P00010,	Total Coliform 10 10 10 10 10 10 10	Alkalinity as CaCo ₃ (40) P00410.
D.O Winkler(3) P00300,	Fecal Streptococci to 1 10 10 10 10 10 10	☐ Min. Acidity ☐ as CaCo ₃ (41) P00436,
. D.O Probe (4) P00299.	Fecal coli	☐ Chioride (42) P00940.
□ PH (Field): (5) P00400,	#100 ml MF (25)P31613	☐ MBAS (43) P38260,
Sample Deptn-ft. (6) P00003,	Fecal Strept (26)P31677	☐ Phenois (44) P32730,
Stream (7) P00061,	MPN/100ml (26)P31677	Hardness - tot (45) P00900,
Gage Height-ft. (8)P00065	☐ Tot coli (27)P31505	☐ Sulfate (46) P00945,
□ Spec. Cond. ● 25°C (9)P00095	MPN/100 mi	☐ Oii & Grease (47) P00556.
Salinity 9,00 (10)900486.	BIOCHEMICAL OXYGEN DEMAND	Petroleum Hydrocarbons(48) P45501.
☐ Tide Stage (11)P70211,	INITIAL D.O. (Iab.) SAMPLE SEED YES NO	☐ Cyanide (49) P00720,
CONDITION CODES	EONC. %	☐ As - tot ug/1 (50)P01002.
Weather Conditions (12) P00041.		Cd - tot ug/i (51)P01027,
Flow Severity (13) P01351.	BOD_	☐ Cr - tot ug/l (52)P01034.
Severity (14) P013,	☐ BOD 5-DAY(28) P310, 6-DAY(29) P312,	☐ Cu - tot ug/l (53) P01042.
Severity (15) P013		☐ Fe - tot ug/l (54) P01045.
	□ coo (30) P340,	BHg - tot ug/ (55) P71900 5K
NUTRIENTS LEVEL DHIGH LOW		Mn fot ug/(56) P01055
NO ₂ - N (16)P00615	□ TOC (5.01) P00680.	☐ Ni - tot ug/i (57) P01067
· · · · · · · · · · · · · · · · · · ·		☐ Pb - tot ug/l (58) P01051
102 + NO3 - N (17)P00630	Color Pt - Cou (32)P00080,	\ \
H ₃ - N (18)P00610,	☐ Turbidity (33)P00076,	☐ Zn - tot ug/l (59) P01092,
t. Kjeldahi N (19)P00625	Suspended Solids (34) P0 0530, Suspended Solids (35) P0 0540,	ADDITIONAL ANALYSIS
P 🔲 (20) P70507,	Ash	P T
PO ₄ □ (21) P00660, □ □ □ ,	☐ Tot. Solids (36)P00500,	P
Prus- ☐ (22) P00665.	☐ Tot. Solids - Ash (37)P00510, ☐ Tot. Dissolved (38)P70300,	D P
b ₄ □ (23) P00650, □ □ □ ,	Solids (TDS)	DP
RESULTS mg/l unless otherwise n	oted	• • • • • • • • • • • • • • • • • • •
11202213	Chemist Review	•
Part 1 (White) - Wate	r Quality Inventory Copy Part 3 (Pink)	- Laboratory Copy

Part 2 (Canary) - Laboratory Copy

Part 3 (Pink) - Laboratory Copy Part 4 (Goldenrod) - Field Samplers Copy

LEY DEPARTMENT OF ENVIRONMENTAL DIVISION OF WATER RESOURCES ENFORCEMENT & REGULATORY SERVICES

ECTION

COMPLIANCE EVALUATION INSPECTION PUBLIC COMMUNITY WATER SUPPLY

DATE	
GENERAL INFORMATION	
PURVEYOR/ MONROE MUNICIPITY UTILITIES ALTHORITY FACILITY	
FILE LOCATION MOINROE TWP PW-ID # 68/1002	
MAILING ADDRESS 37, So. MAIN STREET	
ADMIN. HR J.V. DINOU! CHAIRMAN LICENSES W TO BUSINESS BUSINESS REQUIRED T 2 LICENSES W TO STAV MINCEBACK	
BUSINESS TELEPHONE # Admin.: 609-639-4400 Licensed Operators: T 2 W 2	
FACILITY DESCRIPTION	
SOURCES: descriptions, locations, capacities (mgd): WAS HINGTOW AVE #40.72 snall for limitagence	
presently the will is afflice) Water that Well #5-0.51 mgd. Lake Ave Well #	<u>6-0</u>
Cirkery Lane Well # 7-1.14 mgl.	
Est Tot Eff Cap: 3.27	
TREATMENT: source, type, capacities (mgt). Velle 4,5 in 6 Chlorination (gra) and P H office	tuln
with Caustic Soda. Well # 7 Chlorinetion (gas) and PH adjustment with le	! <u>2000</u> 1
Est Tot Eff Cap: 3.27	
FINISHED WATER STORAGE: descriptions, locations, capacities (mg): Chestrutand Water It served tonk	015
Werkert Blud elevated tank 0.30mg; Carkery Love Black Home Pike elevated	
the same street of the same stre	B:103- (
Est Tot Cap: /, 4/5	
EMERGENCY INTERCONNECTIONS: descriptions, available gallonage(mgd):	·
P. T. A. A. H.	
Est Tot Avail:	- 111
AUXILIARY POWER: location, type, capabilities: Will#4 gaselie lugine direct drive Will#5 engine direct drive, Will#7 divil generator for well a treatment.	jujo
evigent avier war all allow fire it for well & ristingit.	
Tatal 2067	



NJDEP - DIVISION OF WATER RESOURCES PUBLIC COMMUNITY WATER SUPPLY INSPECTION



	DELIVERY INFORMATION	· .	
PLANT DELIVERED WATER (mgd_month_year) Max 36.521 m	of 7/81 - Min 13.12.	Omgal 1/8/ Average 2	7.029
BULK PURCHASES (provider mgd) monte	· · · · · · · · · · · · · · · · · · ·		·
BULK SALES (customer, mgd)			
NUMBER OF SERVICES 3500		% METERED 10	0
MUNICIPALITIES SERVED (est. services in each)	an Monrae Tup		
	7	•	
		TOTAL ESTIMATED POPULATION SERVICED	9400
CURRENT/RECENT WATER RESTRICTIONS Wang		TOTOLATION SERVICES	
NEW CONSTRUCTION	-		·
11000100	(min) to Program	2(max) 45 (max)	
	MONITORING & REPORTIN	<u>G</u>	
PARAMETER(S)	FREQUENCY REQUIRED	FREQUENCY PERFORMED	
Coliform	10 per month	10 fer month	
Inorganics Nitrate	1/3 42	1/3 yr	•
Trihalomethanes Organics	الماريم. ماريم.	N/rt	
Turbidity	NIN	NIH	
RADIOLOGICAL	-114 yw	1144	
NAME OF LABORATORY Quality (entiel Lab	CERTIFICATION #	04002
ADDRESS 243 White House (the Hudubon	mf	
	COMPLIANCE EVALUATIO	<u>N</u> .	
SOURCE DEFICIENCIES Will #4 Men	cury in water Al	west to Gueline Sta	tion.
Will #5 not in operation. C	Put for general or	excitement and repa	inter
Will #5 not in operation. C mercury Contamination belo	w EPA Stanley	Paul Mc Lestellike	Day He Las
Mrenking water Act			
TREATMENT DEFICIENCIES			
·			
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NJDEP - DIVISION OF WATER RESOUR PUBLIC COMMUNITY WATER SUPPLY INSPECTION



COMPLIANCE EVALUATION (Continued)

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ICENSING, MC	INITORING A	AND/OR RE	PORTING D	EFICIENCIES	11000				
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OMPLIANCE S	AMPI ING VI	OI ATIONS	:			······································	 		
LOCATION	DATA SOURCE	PARAM	MAX CONTMNT LEVEL	RESULT	LOCATION	DATA SOURCE	PARAM	MAX CONTMNT LEVEL	RESULT
			LLTLL					LEVEL	
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VERALL COM	PLIANCE RA	TING:							• .
	™ AC	CEPTABLE		□ CONDIT	TIONALLY ACC	EPTABLE	[□ UNACCEP	TABLE
OTICE: YOU A	RE REOUR	FD TO INF	ORM THF N	INFPINW	RITING OF YOU	ID ACTITAL O	OR INTENT	OFD ACTIONS	: TO
СОМРІ	LY WITH N.J.	S.A. 58:12A	1- <i>1 ET SEQ.</i> 1	VIA IMPLEMI	ENTATION OF R	REMEDIAL M	EASURES :	TO CORRECT	THE
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NSPECTOR:	Kuc	ALA WI Signature	ev	PER	SON INTERVIEV	wed:	. 11/2	<u>P. L. D. W. N.</u> Name	
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ei.		Name					Night	Title	
•	FEI	-				Man	a Stheen	intel VI	Ation
		Title				111310	Org	garlization	iare _
		11						-	
		Region							

·	TABUL			OF ENVIRONM	C WATER SUP	TECTION PLY	-7 -Λ	4
Supply <u>killin</u>	yeer a T					County	suice,	les
Municipality	mal 1	100				Date Collect	ed <u>6/2</u>	182
Collected by \	eclaric	" Sega	ion by	Memo #		Project # W-	·	
BACTERIOLOGICAL	. ANALYSES: (Coliform organis Chlorine residual	ms determined b	y the membrane filt	er technique are	reported in c	olonies per	100 ml.
Sample Number	Co. Co. No. 10 Co.						Chlorine	Residual
J. (1/ 2	Sample Number Point of Collection				13504	Total	Free	Total
19640	WW 36	- Jake	Hul.		204	 	 	
14642	10111 3	Wall	v st		+	 	 	
					-	 		
	+							
	+		-		-		 	
						<u> </u>	<u> </u>	1
PHYSICAL - CHEM	ICAL ANALYSE			andards and/or Natio				es are from
Sample Number	14410			Sample Number	14040	14/642	<u>'-</u>	
Color (10)				Arsenic (0.05)		 		
Odor (III)		·		Barium (1.0)		<u> </u>		
Turbidity (5)		<u> </u>		Cadmium (0.010)				
pH				Chromium +6 (0.05)				
Alkalinity to pH 4				Copper (1.0)				
Nitrate as NO ₃ (45)	-		<u> </u>	Cyanide (0.20)				
Chloride (250)				Lead (0.05)	200 TV			
Total Dis. Solids (500)				Mercury (0.002)	=0.0005K	0.0005 K	=	
ABS/LAS (0.5)				Selenium(0.01)				
Total Hardness (150)				Silver (0.05)				
				Phenol (0.001)				
Total Iron (0.3)				Endrin (0.0002)		į		
Total Iron (0.3) Manganese (0.05)			4	II				
				Lindane (0.004)	251	FIV	ED	
Manganese (0.05)		2		Lindane (0.004) Methoxychlor (0.1)		EIN		
Manganese (0.05) Sodium (50)						<u> </u>		
Manganese (0.05) Sodium (50) Sulphate (250)				Methoxychlor (0.1)	il.	N 25198	2	
Manganese (0.05) Sodium (50) Sulphate (250) Fluoride (2.0)				Methoxychlor (0.1) Toxaphene (0.005)	il.	N 25198	2	
Manganese (0.05) Sodium (50) Sulphate (250) Fluoride (2.0)				Methoxychlor (0.1) Toxaphene (0.005) 2.4-D (0.1)	il.	<u> </u>	2	
Manganese (0.05) Sodium (50) Sulphate (250) Fluoride (2.0)				Methoxychlor (0.1) Toxaphene (0.005) 2.4-D (0.1)	L STATE DE	N 25 198	ALL PRITECT!	750
Manganese (0.05) Sodium (50) Sulphate (250) Fluoride (2.0) Zinc (5.0)				Methoxychlor (0.1) Toxaphene (0.005) 2.4-D (0.1) Silvex (0.01)	L STATE DE	N 25198	ALL PRITECT!	TED
Manganese (0.05) Sodium (50) Sulphate (250) Fluoride (2.0)			0 0/24	Methoxychlor (0.1) Toxaphene (0.005) 2.4-D (0.1) Silvex (0.01)	L STATE DE	N 25 198	ALL PRITECT!	TED

1983

Quality			orator:	y an	vironment	al Sup	port Allia	anc		ormation		958
Audubon, N	lew Jerse	y 08106		DEP Lab. I	D# 04002	609-428-1	303 609-29	76-7970	Date rec	'd at lab	4/14/8	3 _T
Customer I	nformati	<u>on</u>		,								
Co	Monroe	M.U.A	•				<u> </u>	_	·			т
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Address _								- [•			
Sample dra	Gប: wn by	⁹ M Da	ste	14/83	Time1 P.M	•		-				т_
	•								Date an	alyses co	mp	т
From	#5	rell						- '	•			
					CH	HEMICA	A L					
Test		Quan.	Meth.	Tech.	Date/Time		Test		Quan.	Meth.	Tech.	Date/Time
Acidity		-	 	<u> </u>			Lead		 -			
Alkalinity		<u> </u>	├	 			Lithium		 	<u> </u>		
Aluminum		 	ļ		 		Magnesium		 	├ ──		
Ammonia		ļ	ļ		 		Manganese			 	 	
Antimony		 		ļ			Mercury		less	than .	002mg/	1
Arsenic		 -	 	ļ	· · · · · · · · · · · · · · · · · · ·		Molybdenu	m	_	 		
Acid Extr.		 -	ļ				Moisture		 	ļ	ļl	
Barium		 					Nickel		ļ			·
BOD (5 day)	_			<u></u>		Nitrate		 			
Bromides		ļ			<u> </u>		Nitrite		 		 	
Bismuth		 	ļ				Odor		<u> </u>	ļ		
Base-Neutr	al Extr.	<u> </u>					Oils & Great		 			
Cyanide		ļ	L				Pesticides 8	PCB's			L	
Chlorine		<u> </u>					Petro Hydro	carbons	<u> </u>			
Chlorides		<u> </u>					Phenols		<u> </u>	<u> </u>	<u> </u>	
Cadmium							Phosphate		<u> </u>			···
Calcium							Phosphorus					
Carbon Dio	kide						Potassium					
Cesium							Platinum					
COD							Selenium					
Chromium							Silver		1			
Cobalt							Sulfate					
Color							Sulfite		1			
Copper		† — — ·					Sulfide		1			
Density							Sodium				1	
D. Oxygen						_	Silica					
Detergents							S. Matter			·		
ABS/LAS							THM		1			
Fluorides							Thallium		1			
Fungicides							Turbidity				1	
Hardness		1					Taste		1		 	,
Hydrogen S	ul	 					T. Solids		 			
Herbicides	<u> </u>						TDS		 			
Iron							TOC		 			
Iodine		t	<u> </u>				Vanadium		 			
lodide									+		 	
Kjeldahi N							Volatile Org.		 			······································
Lanthan							Volatile Solid	15	 		 	
Cammon							Zinc Zirconium		 			
						:			1			
					BACTE	RIOLO	GICAL		-,			
Test	DII.	Total	Meth.	Tech.	Date/Time	-	Test	Dil.	Total	Meth.	Tech.	Date/Time
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T. Collform	<u> </u>	<u> </u>		1		· <u>[</u>	F. Strep		<u> </u>		li	
0										2		
Remarks _										/-/-		0 2

Bruce Greenwald, Lab Mgr.

243 White Horse P		00104	Cerm		Laboratory"	609-428-1303	/00 0-	. 7074		i		A 957
Audubon, New Jersey 08106 Customer Information				DEP Lab. I					Date rec'd at lab 4/14/83_T			
		•	A		Ptoble	2 /1/2	4,2,	.	Ву			
0	Monroe M. J.A. Potable Water								Date rec'd by Tech T			
		:	•			1000			Bv			
ddress						Phone		- [·			
Sample drawn by Gus M Date 4/14/83 Time 1 P.M.								_ 1	Date analyses started T			
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rom	#4	well						- '				
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est	1	Quan.	Meth.	Tech.	-	Tes	t		Quan.	Meth.	Tech.	Date/Time
cidity						Lea	ıd					
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ntimony						Me	rcury		.0046	mg/1		
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ase-Neutral Ext	<u>r.</u> +						& Great		 			
yanide	\dashv						ticides &					
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olor •						Sulf	ite					
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Bruce Greenwald, Lab Mgr.

JUN 20 1984

Monroe Township M.U.A. 372 South Wain Street Williamstown, New Jersey 08094

RE: Monitoring for Mercury
P.W. - ID No.: 0811002
Hoorce Township/Gloucester County

Gentlemen:

The Department's records of Well \$4 indicates that the raw water from this well contains mercury in excess of the MLL established by the Safe Drinking Water Act. It is known that Well \$4 is used for emergency only, investigate and discount to maintain a monthly surveillance of Well \$4 and the test well on Avery Drive for mercury. This monitoring is to be reported to this Department along with the monthly operators reports.

If you have any questions concerning the above, please contact Al Anderson, the Compliance Investigator responsible for this case, who can be reached at (609) 292-1924 or by letter through this Division.

Thank you for your cooperation.

Very truly yours,

Original signed & mailed

Nick DeMao Supervising Environmental Compliance Investigator Southern Region Enforcement Element

A5leral

ce: Ceorge Cossaboon, Licensed Operator Sureau of Potable Water Robert Williams, USEPA - Region II

bcc: Region File THROUGH: DeMeo & Ricciardi

Central Pile Al Anderson Marianne Montgomery

B-37

Quality Contro	!	"Certif	led To.y	Laboratory"	_			rmation #) 695
Audubon, New Jerse			DEP Lab. 10	# 04002	609-428- 1	303 609-296-7970	Date rec	'd at lab	11/26/	′84т 1:45PM
Monm	- M II		•				Ву			
CoMonr	oe M.U	• A					Date rec	d by Tecl	n	т
Address	·				Phone		Ву			
Sample drawn by B.	G. Da	te <u>11</u>	/26/84	Time11 :	33 A.M.		Date ana	lyses stai	rted	T
							Date ana	lyses cor	np	T
FromWasn	ington	Ave.	Well #4	+	·			•		
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Alkalinity					· `···	Lithium				
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Ammonia						Manganese				
Antimony				-		Mercury	.00295	mg/1		
Arsenic						Molybdenum				
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BOD (5 day)					 ,	Nitrate				
Bromides						Nitrite	T			
Bismuth						Odor	1			
Base-Neutral Extr.						Oils & Grease				
Cyanide					 .	Pesticides & PCB's				
Chlorine						Petro Hydrocarbons				
Chlorides					_	Phenois	1			
Cadmium						Phosphate	1			
Calcium						Phosphorus	†			
Carbon Dioxide	1					Potassium				
Cesium				· · · · · · · · · · · · · · · · · · ·		Platinum				
COD						Selenium	1			
Chromium		<u> </u>				Silver	 			
Cobalt	 					Sulfate	 		 	
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Copper	 	<u> </u>	 		- .	Sulfide	+			
Density	 					Sodium	+		 	
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Remarks										
										B-38

Customer Information Co. Monroe M.U.A. Williamstown, N. Sample drawn by B.G. Date 12/17/ From Washington Ave. Test Quan. Meth. To Acidity Alkalinity Aluminum Ammonia Antimony Arsenic Acid Extr. Barium BOD (5 day) Bromides Bismuth Base-Neutral Extr. Cyanide Chlorine Chlorine Chlorine Chlorides Cadmium Calcium Carbon Dioxide Cesium COD Chromium Coobalt Color Copper Density D. Oxygen Detergents ABS/LAS Fluorides Fungicides Hardness Hydrogen Sul. Herbicides Iron Iodine Iodine Iodine Iodide Iodide Kjeldahl N Lanthan	34 Time 11:	CHEMICAL Test Lead Lithium Magnesium Manganese Mercury Molybdenur Moisture Nickel Nitrate Nitrite Odor	By	'd by Tecl	n	т
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Test Dil. Total Meth. Te		ERIOLOGICAL	_			Date/Time
T. Cells			Dil. Total	Meth.	Tech.	Date/ Line
	BACT	F. Coliform		<u></u>	├	
T. Collform		F. Strep	LL	L	<u></u>	

243 White Ho	rse Pike	:			'nvironmer	ntal Support			Lab. Info	ermation #D_2	320	
Audubon, Ne	w Jerse	y 08106		DEP Lab. 10	O# 04002	609-428-1303	609-296	6-7970	Date rec	'd at laß./	24/85	T2:30PM
Customer In				,					Ву			
Co	Monro	oe M.U	.A.					. [Date rec	d by Tech	ı	т
	Wilia	amstowi	n N T					- 1				
Address						Phone		-:				
Sample draw	nby B.	.G. Da	te 1/24	4/85	Time	1:51 P.M.		. 1	Date ana	lyses star	ted	т
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				"		-/-	1	7	- /	/2 .	<i>i</i> '	
									Te	<u>(B</u>	ruce Gre	enwald, Lab Mgr

Quality (243 White Ho Audubon, Ne	rse Pike	!	"Certif		Labo		tal Support Alliance 609-428-1303 609-296-79			E 904	5 /85 2:50pm
Customer In	iormatic	<u>on</u>					1				
Co	Monro	e Town	ship M	Ú <u>A</u>				•			т
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Alkalinity Aluminum		 		 	┼-		Magnesium	 	<u> </u>		
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BOD (5 day)				 	┼		Nitrate			 	
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DEPARTMENT OF ENVIRONMENTAL P **DIVISION OF WATER RESOURCES ENFORCEMENT & REGULATORY SERVICES**

LCTION



COMPLIANCE EVALUATION INSPECTION PUBLIC COMMUNITY WATER SUPPLY

GENERAL INFORMATI	ON
PURVEYOR/ MONROE TOWNShip M.U.	A. The second se
FILE LOCATION MONROE Tup. / Coloreste	R CO. PW.ID # 0811002
MAILING ADDRESS 372 South Main ST.	reet, Williamstown, N.J.
ADMIN MS Jacqueline Schoenewal	REQUIRED T-3 LICENSES W-3 GEORGE
BUSINESS TELEPHONE # Admin.: 629-7586 Licensed Operators:	T-3 C-3 W-4 M. CossaborE
FACILITY DESCRIPTION	트 프로 근처 함께 제한 취임 및 및 트립스템 클럽스템 그는 그는 그녀들들도 됩니다. 그 모고 있습니다 모습으로 드라고
SOURCES: descriptions, locations, capacities(mgd): 101/1#4 W	3
Well#5 Water St. 0.81 mgd; Well #	b Lake Ave 0.60 mgd;
Well #7 Corkery LANE 1.14 mgd.	
Well #4 emergency use only	Est Tot Eff Cap: 3.27 nigd
TREATMENT: source, type, capacities(mgd):	VATION AT All wells.
wells 4,5 ; 6 PH najusted with	
PH adjusted with line.	
	Est Tot Eff Cap: 3.27 mgd
FINISHED WATER STORAGE: descriptions, locations, capacities(mg):	voted Truk-Water ST.
0.15 mg; Elevated Tank-Herbert	Boulevard 0,30 mg;
Elevated Tank-Corkery LANE 1.0	をたった (A) ちょうしょ かいしょう サー・コースカー・ケー・フェー・フェンス・コンス あっちょう
	Est Tot Cap: 1:413 mg
TARROCENCY INTERCONNECTIONS: 4	lime
EMERGENCY INTERCONNECTIONS: descriptions, available gallonage(mgd): _	
	Est Tot Avail:
AUXILIARY POWER: location, type, capabilities: WEll 45 (Kig)	of Angle Unive) Olose
engine/generator for well and	Trestment-50 Kw- 0.81 mgC;
Well #7 diesel enjue/generator	for well and
Treatment - 10.0 Km - 1.14 mgd.	HI AUX CAP 1.95 MAS
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NJDEP - DIVISION OF WATER RESOURCES PUBLIC COMMUNITY WATER SUPPLY INSPECTION



Page 3

COMPLIANCE EVALUATION (Continued) TORAGE AND/OR DISTRIBUTION DEFICIENCIES LICENSING, MONITORING AND/OR REPORTING DEFICIENCIES. **COMPLIANCE SAMPLING VIOLATIONS:** MAX HAX DATA DATA LOCATION. CONTMNT CONTMNT PARAM RESULT LOCATION PARAM RESULT **SOURCE** SOURCE LEYEL LEYEL ---..... 1 OVERALL COMPLIANCE RATING: M ACCEPTABLE CONDITIONALLY ACCEPTABLE UNACCEPTABLE NOTICE: YOU ARE REQUIRED TO INFORM THE N.J.D.E.P. IN WRITING OF YOUR ACTUAL OR INTENDED ACTIONS TO COMPLY WITH N.J.S.A. 58:12A-1 ET SEQ. VIA IMPLEMENTATION OF REMEDIAL MEASURES TO CORRECT THE DEFICIENCIES LISTED IN THIS REPORT. FAILURE TO ADEQUATELY RESPOND IN A TIMELY FASHION WILL RENDER YOU LIABLE FOR PENALTIES OF UP TO \$5,000.00 FOR EACH VIOLATION, PURSUANT TO N.J.A.C. 7:10-3. INSPECTOR: PERSON INTERVIEWED: Signature



PUBLIC COMMUNITY WATER SUPPLY INSPECTION



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	Pressures	5 05) (min) to	65 ps; (max)	
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	Coliform	13- month	15- nunth	
	Inorganics NA	1- 3 YIMS	1-3 yes 10-1-84	
	1/20140	1-3 YEARS	1-3 yps	
	Trihalomethanes	+	4-3 nio's 5-30-84	
	Organics Turbidity	NA		
	Radio	1-4 YIARS	1-4 yrs 6-24-83	Completed
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	1-280'5	1- 6 mo's	1-6 mo's 5-30-84	
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DEPARTMENT OF ENVIRONMENTAL NEW J DIVISION OF WATER RESOURCES

ENFORCEMENT & REGULATORY SERVICES



COMPLIANCE EVALUATION INSPECTION PUBLIC COMMUNITY WATER SUPPLY

	-	/ _

1986	FORLI	C COMMUNITY	WATER SUPPL	DATE	JAN 5 1756
		GENERAL INFO	RMATION		
PURVEYOR/ FACILITY	CNACE TU	12 176	1-7		
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	CCCEUNE SO	HURNURL		REQUIRED T	- 3 - 3
BUSINESS TELEPHONE # Adm	99 - 2586 in.:			SSHBLNE	W 6. COSSIDERINE
		FACILITY DESC		# (USIS CNIN
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0-30 ME .	SEVETED	TANK	C.C.A.M.	KY 2	ONT 10 MG
) ·				Est Tot Cap: .	1-45 196
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AUXILIARY POWER: 1	oçation, type, capabilities:	1154 #5	DIESE	7 Ris.	WT FINGLE
DIZIUE 32	DRUE EVE	2 1200	1931	(.61)	<u> </u>
WELL #	DIESTE O	SENELAT	CR.	SOURCE	- suo
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JOEP - DIVISION OF WATER RESOURCE PUBLIC COMMUNITY WATER SUPPLY INSPECTION



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	Inorganics & NA	EVERY 3 YRS	10/1/89		-
	Nitrate	EVERY 3 XRS	5/30/59	<u> </u>	-
	Trihalomethanes	4-57165	9-3145 5		-
	Organics			<u> </u>	-
		//4	1/4		→
	Turbidity	N/A	1/4		 }
	KADICNUCLIDES	EURRY A YRS	6/64		-
	STRENDALY KEES		10/04		<u>-</u>
	L'EEC	JAN - SUNE	1/65-5	12.5	_
	Caranton174	LIVES CNEY	10/84		_]
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NJDEP - DIVISION OF WATER RESOURCE PUBLIC COMMUNITY WATER SUPPLY INSPECTION



COMPLIANCE EVALUATION (Continued)

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						7 193						

BUREAU OF POTABLE WATER

DISPOSITION: NPA

Supervisor Signature & Date

Investigator Signature

ATTACHMENT



COUNTY OF GLOUCESTER STATE OF NEW JERSEY DEPARTMENT OF HEALTH CARPENTER ST. & ALLENS LANE

WOODBURY, NEW JERSEY 08096-2699 (609) 853-3405

ROY L. BAYLOR FREEHOLDER

ROBERT J. SMITH, M.P.H., DIRECTOR DEPARTMENT OF HEALTH

Monroe Township Board of Health Environmental Services Building Academy Street Williamstown, NJ 08094 September 3, 1986

SUBJECT:

Complaint #86364.

Individual Water Supply System Environmental Services Buildin

Monroe Township

Enclosed is a copy of the water analysis results for the sample collected from the subject water supply on July 17, 1986. The sample was analyzed for heavy metals. The results confirm the previously found mercury concentration. The source of the mercury concentration is unknown. During our investigation the cleaning products stored in the building were noted and none indicated that they contained mercury. Additionally the company which distributes the weed killer (Sunshine Chemical) present was contacted and they confirmed that no mercury was contained in those products either.

In interviews with office personnel it was indicated that the area was an orchard in the past. It is possible the mercury originated in sprays used at that time which have now reached the water table.

Due to the excessive mercury concentration present the water should not be used for drinking or cooking. It is recommended that bottled water be provided until a permanent alternate supply can be supplied.

If you have any questions, please do not hesitate to contact this office.

Very sincerely yours,

STEVEN WEBER

Chief Sanitary Inspector

SW/aal Enclosure

1686 3 1986

Three Hundred Years of Public Service



Gloucester County Dept. of Health August 7, 1986 Page 2 of 2

I. Methodology

This analysis adhered to the methods described in:

. EPA Manual of Methods for the Analysis of Water and Wastes, 1979.

II. Analytical Results

A. Metals and Bacteria

Sample Designation

Parameter	NAC1487 SW860717-A	
Arsenic, total, mg/l Barium, total, mg/l Cadmium, total, mg/l Chromium, total, mg/l Lead, total, mg/l Mercury, total, mg/l Selenium, total, mg/l Silver, total, mg/l Sodium, total, mg/l Copper, total, mg/l Iron, total, mg/l Manganese, total, mg/l Zinc, total, mg/l	<0.01 <0.10 <0.01 <0.05 <0.05 <0.01 <0.01 <0.01 7.5 0.23 0.19 0.042 <0.05	mux allowable comming

18 E. High Street Glassboro, New Jersey 08028

881-8229

BORATORY

Water Analysis Field Data Sheet

Municipality					
Municipality Monroe	Reason for sample (e.g. new well)				
Owner Bd A Education	Phone number				
Location Control of Tan	Mailing Address				
Env. Services B/ds	Neadenin St Winstown				
Sample # Time of Collection	Date sampled Name of sampler				
W860717A 2:00pm	7/17/86 S. Weller				
Diagram of sample Sites (non-potable)	Sample taken from Water Temp.				
	Outside faucet				
	Depth of Well 7 'Shallow Well to Septic (type)				
:	Analysis requested 10920				
	Mercury + Metals				
Field Comments, sampling conditions	Received at lab Lab #				
for water 10 min	Date: W D Analysis conducted				
	Time: Date:				
	Time: 7-21-86 Analysis conducted Date: Time:				
LABORATO	DRY RESULTS				
BACTERIOLOGICAL	CHEMICAL				
A. MEMBRANE FILTER (colonies/100 ml.)	1. Nitrate (NO ₃ -N mg/l)				
Total Coliforms:	2. pH				
Fecal Coliforms:	3. Iron (mg/l)				
Fecal Streptococci:	4. Manganese (mg/l)				
B. MPN (most probable number/100 ml)	5. M.B.A.S. (mg/1 LAS)				
Total Coliforms:					
Fecal Coliforms:					
Fecal Streptococci:					

Laboratory Director

C. Total Plate Count (colonies/ml)

DESCRIPTION Monroe Sump

NJ005359

DATE

MAY 15 1984

PLEASE RETURN TO DEBBIE OCHANAS, BGWDP, AFTER REVIEWED/SIGNED, ETC. THANK YOU.

JOHN W. GASTON JR., P.E. DIRECTOR

MEMORANDUM

TO:

Edward Londres, Assistant Director

Division of Waste Management

FROM:

John J. Trela, Chief

Bureau of Ground Water Discharge Permits

SUBJECT:

Monroe Township Sanitary Landfill

NJPDES Permit No. 0053597

Attached please find for your review a draft permit to monitor ground water discharges for the above cited landfill, pursuant to NJPDES regulations.

We will issue the NJPDES permit according to regulations after we have received and reviewed your comments and recommendations within 10 working days of the date of this memo. If comments are not returned within 10 working days, the permit will be issued for public notice in its present form.

WOM111

Attachments

cc: (with attachments)

George McCann Paul Kurisko Joseph Rogalski

MONROE TWP MUNICIPAL WELL #4 & #5 MONROE TWP/GLOUCESTER COUNTY NEW JERSEY EPA # NJD980769699

REFERENCE:

I. MAPS

- 1. USGS QUADRANGLE MAP
- 2. COUNTY MAP
- 3. STATE ATLAS MAP
- 4. STATE ATLAS WATER SUPPLY MAP
- 5. MONROE MUA WATER SYSTEM
- 6. TAX MAP WELL #4
- 7. TAX MAP WELL #5
- 8. TAX MAP POTENTIAL SOURCES

II. ATTACHMENTS:

- A. WELL RECORDS NJDEP
- B. HISTORICAL SITE INFORMATION 1976 1986
- C. GLOUCESTER COUNTY DEPT. OF HEALTH INVESTIGATION
- D. MONROE TWP SANITARY LANDFILL
- E. DANGEROUS PROPERTIES OF INDUSTRIAL MATERIALS IRVING SAX
- F. THE MERCK INDEX
- G. PESTICIDES IN THE ENVIRONMENT
- H. MEMO TO FILE FROM FRANK FARANCA



STATE OF NEW JERS! > IMENT OF ENVIRONMENTAL PROTECT CN 402

Trenton, N.J. 08625



		PERMIT					
The New Jessey Department of Environmental Protection grants this permit in accordance with your application, attachments accompanying same application, and applicable laws and regulations. This permit is also subject to the further conditions and stipulations enumerated in the supporting documents which are agreed to by the permittee upon acceptance of the permit.							
Permit No. NJ0053597	Issuance Date	·	Effective Date	:	Expiration Dat		
Name and Address of Applicant Monroe Township 266 S. Main St. Williamstown, N.J. 080	Ņ	Location of Activity/F Monroe Township Sicklerville Ro Villiamstown, N	Sanitary Landfill	Ţ	d Address of Ow	1	
Issuing Division Water Resources	N	Type of Permit NJPDES Permit for Ground Water	or Discharge to	Statute(s 58:10A N.J.S. 7:14A	N.J.S.A. 1 et seq. 1 et seq.	Application No.	
This permit requires Mon Williamstown by operat specific and general of NJPDES permit is intennamed facility. This current status and impit intended to be an a effects the environment or welfare. The issuance of this idetermination of the the shall not be construed of limited duration.	ing and material interchaical and material interchains and material interchains and material and material interchaical and material and material interchaical and material and	of this initial ablish an adequation of this initial ablish an adequation on a second or surface waterim permit do adequacy of the	ound water monitor I interim NJPDES p uate ground water i to obtain ground w ground water. It that the permittee er quality, or thr es not indicate th information avail	ing wel ermit. monitor ater da shall n has co eatens at the able.	ls accordin The initia ing program ta to evalu ot be const nducted whi the public Department Interim ini	g to the l interim at the above ate the rued, nor is ch adversely health, safety has made a tial permits	
The data generated through the initial interim NJPDES permit will be used by the Department to evaluate the current status and impact of existing facilities on ground water quality. It will also give the Department information to determine if there is any potential or actual threat to public health or safety or damage to the environment due to current or past practices. Based on the information generated by the issuance of this permit, the Department may require the permittee to reduce the quantity of discharge, upgrade or install additional treatment, install additional monitor wells, conduct ground water decontamination procedures or cease discharges to waters of the state.							
The issuance of this'i permit, nor does it respectified in Chapters	lieve the	permittee of the	he duty to submit	additio	nal informa	tion as	

Documents attached hereto shall become part of this permit.

Approved by the Department of Environmental Protection BY AUTHORITY OF:		
JOHN W. GASTON, JR., P.E. DIRECTOR, DIVISION OF WATER RESOURCES	ARNOLD SCHIFFMAN, ADMINISTRATOR WATER QUALITY MANAGEMENT	DATE

or as may be required by the Department prior to permit renewal. Additionally, this initial interim NJPDES permit does not relieve the permittee of any liabilities associated with public

health or safety problems or environmental damage created as a result of the permittee's

activities.

